

1 **Microencapsulation of phenolic compounds from olive pomace using spray drying: a study of**
2 **operative parameters**

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4 Marco Pains^{a,b*}, Bahar Aliakbarian^{a,b}, Alessandro A. Casazza^{a,b}, Alberto Lagazzo^a, Rodolfo Botter^a
5 and Patrizia Perego^{a,b}
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9 *^aDepartment of Civil, Chemical and Environmental Engineering, University of Genoa, via Opera Pia*
10 *15, 16145 Genoa, Italy*

11
12 *^bResearch Center for Biologically Inspired Engineering in Vascular Medicine and Longevity*
13 *(BELONG), Via Montallegro 1,16145 Genoa, Italy*
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25 *Corresponding author. Tel.: +39 010 3532584; fax: +39 010 3532586.

26 *E-mail address:* marco.paini@unige.it (Marco Pains)

27 **Abbreviations**

28 ARP: Antiradical Power, CAE: Caffeic Acid Equivalents, DM: Dry Matter, DP: Dried Powder,
29 DPPH': 2,2-diphenyl-1-picrylhydrazyl, DTA: Differential Thermal Analysis, DTG: Differential
30 Thermogravimetric analysis, HPTE: High Pressure and Temperature Extraction, IT: Inlet
31 Temperature, IT130: Inlet Temperature 130 °C, IT160: Inlet Temperature 160 °C, MD:
32 Maltodextrins, MD10: Maltodextrins 100 g/L, MD50: Maltodextrins 500 g/L, ME:
33 Microencapsulation Efficiency, OT: Outlet Temperature, OP: olive Pomace; OPE: Olive Pomace
34 Extract, SC: Swelling Capacity, SEM: Scanning Electron Microscopy, sp: surface polyphenols, SP:
35 Surface polyphenol Percentage, TG: Thermogravimetric analysis, TP: Total Polyphenols, WAI:
36 Water Absorption Index, WSI: Water Solubility Index.

37

38 **Abstract**

39 In this work, polyphenols from olive pomace were extracted with a high pressure-high temperature
40 agitated reactor and for the first time encapsulated by spray drying, using maltodextrin as coating
41 agent at different concentrations. In addition, effects of inlet temperature (130 °C and 160 °C) and
42 feed flow (5 mL/min and 10 mL/min) were studied. Physicochemical, antioxidant properties and
43 stability of microparticles were evaluated. High inlet temperature implied lower moisture and bulk
44 density without affecting antioxidant properties. Increasing maltodextrin concentration caused lower
45 bulk density and higher microparticles sizes, while higher feed flow lead to increased moisture
46 content. Spray drying at inlet temperature of 130 °C, maltodextrin concentration of 100 g/L and feed
47 flow of 10 mL/min, resulted in high microencapsulation yield (94%) and encapsulation efficiency
48 (76%) with high polyphenols content (39.5 mg_{CAE}/g_{DP}) and antiradical power (33.8
49 mmol_{DPPH'}/L_{extract}). HPLC and thermogravimetric analysis revealed the thermal protection effect of
50 maltodextrin for phenolic compounds. Microcapsules were stable at 5 °C in dark condition for 70
51 days, and only 21% were degraded increasing storage temperature up to 25 °C. UV light exposure
52 resulted in a 66% loss in polyphenols after 48 hours of exposure.

53

54 **Keywords:** polyphenols; microparticles stability; antiradical power; hydration properties; stability

55 test.

56

57 **1. Introduction**

58 Olive pomace (OP), as the main by-product of the olive oil production, is currently treated as an
59 industrial waste or used as combustible material, nonetheless this organic matrix can have potential
60 applications in compost production, heavy-metal absorption or biofuels feedstock (Che,
61 Sarantopoulos, Tsoutsos, & Gekas, 2012; Haddadin, Haddadin, Arabiyat, & Hattar, 2009; Mavros,
62 Xekoukoulotakis, Mantzavinos, & Diamadopoulos, 2008; Michailides, Christou, Akkratos,
63 Tekerlekopoulou, & Vayenas, 2011; Miranda et al., 2012; Pagnanelli, Mainelli, Veglio, & Toro,
64 2003; Tekin & Dalgic, 2000). Moreover, OP is an interesting waste containing polyphenols
65 (Aliakbarian, Casazza, & Perego, 2011; Aliakbarian, Palmieri, Casazza, Palombo, & Perego, 2012;
66 Cioffi et al., 2010), which are bioactive molecules with promising properties (Casazza, Aliakbarian,
67 De Faveri, Fiori, & Perego, 2012; Suarez, Romero, Ramo, Macia, & Motilva, 2009; Tuck & Hayball,
68 2002), protecting living systems from severe diseases such as cardiovascular dysfunctions
69 (Aliakbarian et al., 2012; Palmieri et al., 2012). Owing to these properties, polyphenols have been
70 considered a promising source of antioxidants, which can be used in pharmaceutical, cosmetic and
71 food industries. Recovering these compounds from agri-food residues has been recognized as a big
72 scientific effort in the past decade (An, Wilhelm, & Searcy, 2011; Champagne, 2007; Makris,
73 Boskou, & Andrikopoulos, 2007). Despite having these benefits, polyphenols show low water
74 solubility and low stability to environmental conditions (exposure to light, oxygen, temperature and
75 enzymatic activities). For these reasons, microencapsulation systems were studied in order to preserve
76 the biochemical functionalities of the components (de Vos, Faas, Spasojevic, & Sikkema, 2010; Fang
77 & Bhandari, 2012).

78 Spray drying, a common industrial encapsulation technique, has been used in food industry for the
79 preparation of food additives and flavors (Desai & Park, 2005), and consists in the atomization of a
80 liquid product in a hot gas current, generating a powdered product (Gharsallaoui, Roudaut, Chambin,
81 Voilley, & Saurel, 2007). Due to the decrease of the water activity, spray drying ensures
82 microbiological stability, avoids degradation processes, reduces storage and transport costs and

83 enhances instantaneous solubility of the final product (Gharsallaoui et al., 2007). Several studies were
84 performed using spray drying as microencapsulation technology for antioxidant extracts from
85 different plant sources (Fang & Bhandari, 2012; Fazaeli, Emam-Djomeh, Ashtari, & Omid, 2012;
86 Kha, Nguyen, & Roach, 2010; Robert et al., 2010). However, to the best of our knowledge,
87 microencapsulation of polyphenols extracted from OP using spray drying has not been performed yet.
88 The aim of this study was to assess the efficiency of spray drying to microencapsulate phenolic
89 compounds from OP. The extract, characterized in our previous studies, (Aliakbarian et al., 2012;
90 Palmieri et al., 2012), was obtained through high pressure-high temperature extraction (HPTE), which
91 has been recognized as an efficient extraction technology to recover phenolic compounds from
92 different matrices (Aliakbarian et al., 2011, Ben Hamissa et al., 2012; Casazza et al., 2010). Based
93 on its low viscosity at high solid contents, good solubility and notable heat protection capacity
94 (Gharsallaoui et al., 2007; Kha et al., 2010; Robert et al., 2010), maltodextrin has been selected as
95 coating agent. Effects of inlet temperature, feed flow and maltodextrin concentration were
96 investigated, and the physico-chemical properties and encapsulation efficiencies of the powders were
97 evaluated.

98

99 **2. Materials and methods**

100 *2.1 Chemicals*

101 Methanol, acetic acid, ethanol, *n*-hexane, acetonitrile, maltodextrin, Folin–Ciocalteu and 2,2-
102 diphenyl-1-picrylhydrazyl (DPPH') reagents and polyphenols standards were purchased from Sigma-
103 Aldrich (St. Louis, MO, USA). Standard solutions were prepared with methanol and stored in dark
104 bottles at -20 °C.

105

106 *2.2 Polyphenols extraction*

107 OP of Taggiasca cultivar was supplied by an Italian olive oil production plant (Raineri S.p.a.,
108 Chiusanico, Italy) and stored at -20 °C prior to use.

109 Before extraction, OP was pre-treated as described by Aliakbarian et al. (2011). The extraction was
110 performed in a high pressure-high temperature agitated reactor (model 4560, PARR Instrument
111 Company, Moline, USA) using ethanol/water 50:50 as solvent, with other parameters set on the best
112 conditions studied by Aliakbarian et al. (2011). Extracts were centrifuged at 6000xg for 10 min (ALC
113 PK131, Alberta, Canada), filtered (0.22 µm) and stored at 4 °C prior to experiments.

114

115 *2.3 Encapsulation of phenolic extract*

116 Spray drying of OP extract (OPE) was performed using maltodextrins (MD) with 16.5-19.5 dextrose
117 equivalents as coating agent. Fifty milliliters of OPE was mixed with MD until complete
118 homogenization, and the solutions were fed to a Lab Plant SD-04 spray dryer (Huddersfield, UK)
119 with air flow of 30 m³/h and atomization pressure set at the minimum of the instrument. Spray dryer
120 was run with water for 10 min before and after each experiment. In this work two inlet temperatures
121 (IT) (130 °C and 160 °C, named IT130 and IT160, respectively), two MD concentrations (100 g/L
122 and 500 g/L and named MD10 and MD50, respectively) and two feed flows (5 mL/min and 10
123 mL/min) were studied. For each sample, the outlet temperature (OT) was registered.

124 Preliminary studies performed in our laboratory (data not shown) indicated that an IT lower than 130
125 °C causes the formation of a wet film on the walls of the chamber, while an IT higher than 160 °C
126 leads to an excessive polyphenols degradation (da Costa, Barbosa, do Nascimento, & Macedo, 2002).
127 MD concentrations and feed flow ranges were selected based on literature (Fang & Bhandari, 2012;
128 Fazaeli et al., 2012; Kha et al., 2010; Robert et al., 2010). Powders were stored at 4 °C in closed dark
129 vessels before analysis.

130

131 *2.4 Powder characterization*

132 *2.4.1 Moisture content, bulk density and microstructure of particles*

133 Microparticles were dried at 105 °C until a constant weight, and moisture content was calculated
134 based on the loss in weight between before and after drying.

135 Bulk density (g/mL) was determined as described by Fazaeli et al. (2012) using 1 g of powders. The
136 ratio of mass of microparticles and volume occupied determines the bulk density value.
137 Particles microstructure were evaluated with scanning electron microscopy (SEM 515, Philips,
138 Netherlands): small amounts of powders were coated with a 30 nm-thick gold layer and observed in
139 secondary electrons (5.0 kV) at 500x magnification. Diameters of particles were measured on SEM
140 images using Sigma Scan Pro 5 software (Systat Software Inc., USA) and were divided in 11 different
141 size ranges. For each range, the percentage of distribution was calculated considering the number of
142 particles with diameter in that range compared to the total number of particles.

143 144 2.4.2 Water solubility index, water absorption index and swelling capacity

145 Water solubility index (WSI) and water absorption index (WAI) were determined according to
146 Ahmed, Akter, Lee, and Eun (2010), by dissolving 1 g of product in 12 mL of water. WSI and WAI
147 were calculated according to equations 1 and 2, respectively.

$$148 \quad WSI = DW_{sup}/DW_{part} \times 100 \quad (1)$$

$$149 \quad WAI = PW/DW_{part} \quad (2)$$

150
151 where DW_{sup} is the dry weight of the supernatant, DW_{part} is the initial weight of microparticles (dry
152 basis) and PW is the weight of pellet after centrifugation.

153 The swelling capacity (SC) was calculated according to Lai and Cheng (2004) using equation 3:

$$154 \quad SC = DW_{sup}/[DW_{part} \times (100 - WSI)] \quad (3)$$

155 156 2.4.3 Thermogravimetric analysis

157 Thermogravimetric (TG) analysis were carried-out using a TG-DSC Netzsch Gerätebau STA 409
158 (Germany), equipped with a Netzsch410 furnace temperature controller. About 20 mg of
159 microparticles were placed in an alumina crucible and introduced inside the furnace. The test consists

160 in two sections: a dynamic segment, from room temperature to 750 °C at a nominal rate of 10 °C/min
161 in a nitrogen flux of 40 mL/min, and in an isotherm section at 750 °C for 1 h, in an equivalent air
162 flow, in order to measure the weight loss due to the combustion of the sample and the amount of the
163 residual ashes.

164

165 *2.5 Analytical methods*

166 Polyphenols were extracted from microparticles according to Robert et al. (2010) and filtered (0.22
167 µm) prior to analysis.

168 Total polyphenols (TP) content was evaluated using the Folin–Ciocalteu assay, as described by
169 Aliakbarian et al. (2011), using a UV–Vis spectrophotometer (Perkin Elmer, Wellesley, USA) at a
170 wavelength of 725 nm. TP was expressed as milligrams of caffeic acid equivalents (CAE) per gram
171 of dried powder (DP) ($\text{mg}_{\text{CAE}}/\text{g}_{\text{DP}}$).

172 The antiradical power (ARP) of samples were evaluated with the radical scavenging DPPH[·] method,
173 modified by Casazza et al. (2012). TP and ARP of OPE were measured using the same methods.

174 OPE and extracts from microparticles were also characterized using HPLC-DAD (Hewlett Packard,
175 1100 Series, Palo Alto, CA, USA), equipped with a C18 reverse-phase column (Vydac 201TP54,
176 Hesperia, USA) as described by Aliakbarian et al. (2011). The concentration of each phenolic
177 compound was calculated based on its standard solutions and the results were expressed as percentage
178 ratio between single polyphenol contents and the total amount of identified polyphenols.

179

180 *2.6 Microencapsulation yield and microencapsulation efficiency*

181 The microencapsulation yield was calculated as the percentage of TP in the dried powder compared
182 to the TP of OPE. This parameter is useful to evaluate the degradation of phenolic compounds during
183 spray drying.

184 Surface polyphenols percentages (SP) and microencapsulation efficiencies (ME) of microparticles
185 were calculated according to Robert et al. (2010), using equations 4 and 5, respectively:

186 $SP = sp/TP \times 100$ (4)

187 $ME = 100 - SP$ (5)

188

189 *2.7 Stability tests*

190 Microparticles with highest microencapsulation yield and ME were stored at three different
191 temperatures (5 °C, 25 °C and 45 °C) and at three different light conditions (dark, sunlight and
192 artificial light, ensured by a 62 Watt lamp at 2.1 klux) at room temperature for 70 days. Light intensity
193 was measured using a Lutron LX-107 light meter (Lutron LDV S.r.l, Milan, Italy). Samples were also
194 exposed to UV radiation (280-315 nm), generated from a 20 W UV-B lamp (TL 12, Philips, Holland)
195 at room temperature. During storage tests, microparticles were kept in sealed containers under dry
196 conditions to avoid the effects of humidity on the stability of polyphenols. TP and ARP of samples
197 were analyzed at different storage times.

198

199 *2.8 Statistical analysis*

200 All the experiments were carried out in triplicate ($n = 3$), except the tests on the bulk density, that were
201 repeated five times. ANOVA and Tukey's post hoc test ($p < 0.05$) were used ("Statistica 10.0",
202 StatSoft, Tulsa, USA).

203

204 **3. Results and discussion**

205 *3.1 OP extract*

206 The OPE obtained by HPTE was characterized in terms of TP and ARP. The phenolic content of OPE
207 was equal to 57.7 ± 0.5 mg_{CAE}/g_{DM}, higher than the one obtained by Aliakbarian et al. (2011) using the
208 same raw material and methanol as solvent (45.2 mg_{CAE}/g_{DM}).

209 ARP of the extract was equal to 26.9 ± 2.5 mmol_{DPPH}/L_{extract}, lower than the methanolic extract
210 (40.1 ± 1.5 mmol_{DPPH}/L_{extract}) obtained by Aliakbarian et al. (2011). This can be explained considering
211 that different solvents lead to different phenolic profiles: for example, HPTE with methanol shows

212 the presence of hydroxytyrosol and oleuropein, polyphenols with strong antioxidant properties, which
213 were not detected using ethanol/water 50:50 as solvent.

214

215 *3.2 Powder characterization*

216 *3.2.1 Moisture content, bulk density and microstructure of microparticles*

217 The effect of IT, MD concentration and feed flow rate on OT of the process, moisture content and
218 bulk density of microparticles are reported in **Table 1**.

219

Table 1

220 All the OTs were comprised between 75 ± 1 °C and 102 ± 1 °C. The highest OT was obtained for the
221 sample in which the driving forces of evaporation were greater (IT160 and lower feed flow), while
222 the lowest OT was obtained in the sample at IT130 and higher feed flow.

223 For all the samples, except for the ones processed at IT130 and MD50, the increase in feed flow
224 results in a significant decrease of OT ($p < 0.05$). This can be explained considering that a higher
225 amount of injected liquid implies a higher heat transfer from hot air to droplets, resulting in a lower
226 OT (Tan, Ibrahim, Kamil, & Taip, 2011).

227 The moisture content of the products decreased at IT160 ($p < 0.05$). This behavior is similar to that
228 found by Fazaeli et al. (2012) in the spray drying of black mulberry juice and by Kha et al. (2010) in
229 the spray drying of Gac fruit extract (with a decrease in the moisture content from 5.3% to 3.9% by
230 increasing IT from 120 to 200 °C), confirming that higher ITs enhance the heat transfer between air
231 and droplets promoting moisture evaporation.

232 At IT160, no significant differences ($p < 0.05$) were observed for samples obtained at different MD
233 concentrations and feed flows. At IT130 and MD10, the moisture content increased changing the feed
234 flow from 5 to 10 mL/min, confirming that increasing solution injection causes a less efficient
235 removal of the solvent, and consequently lower OTs and higher moisture contents.

236 As shown in **Table 1**, an increase in IT and MD concentration is associated to a decrease in the bulk
237 density of the powders (e.g. for samples spray dried at MD50 and feed flow 5 mL/min, the bulk
238 density passed from 0.13 g/mL to 0.04 g/mL increasing IT).

239 These parameters also influence the microstructure of the particles (**Fig. 1**). In particular, IT160
240 causes a higher degradation of the spherical structure of the particles (**Fig. 1a-c** and **Fig. 1b-d**), which
241 leads to a higher porosity in the product, and consequently to a lower bulk density of the
242 microparticles.

243 In all the samples, aggregation of the microparticles has been noted, similarly to Fazaeli et al. (2012)
244 in the spray drying of black mulberry juice, in which this phenomenon is related to low glass transition
245 temperatures of the coating agents, which can be easily overcome by OT.

246 **Fig.1**

247 **Fig. 2** shows the percentage of particles with a diameter comprised in 11 different size ranges.

248 **Fig.2**

249 An increase in MD concentration leads to an increasing size. This is particularly evident for the
250 samples spray dried at IT130 and feed flow at 10 mL/min: at MD10 (**Fig. 2a**), 99% of the
251 microcapsules show a diameter in the range 0-20 μm , while at MD50 (**Fig. 2b**), 53% of the diameters
252 are in the range 20-50 μm , 19% are in the range 50-100 μm , and only 29% are in the range 0-20 μm .

253 This difference is evident also comparing SEM images of the two samples (**Fig. 1a-b**).

254 The operative conditions that generate a smaller microparticles size were MD10 and feed flow 10
255 mL/min, with 99% and 97% of the particles in the range 0-20 μm for IT130 and IT160, respectively.

256 For all the samples, average diameters are smaller than 100 μm , with a high percentage under 50 μm .

257 These results are similar to those obtained by Ersus and Yurdagel (2007) operating at IT between 160
258 and 200 $^{\circ}\text{C}$, feed flow rate at 5 mL/min and MD concentration of 6% solid content, but higher than
259 those obtained by Fazaeli et al. (2012), operating at IT between 110 and 150 $^{\circ}\text{C}$, feed flow rate at 150
260 mL/h and carrier agent concentrations between 1.5-16% (w/w).

261

262 3.2.2 WSI, WAI and swelling capacity

263 Rehydration properties of microparticles were evaluated in terms of WSI, WAI and SC. As can be
264 seen in **Table 2**, for all the samples WSI is comprised in the range 79-87%.

265 **Table 2**

266 These values are similar to those calculated by Abadio, Domingues, Borges, and Oliveira (2004) for
267 the spray drying of pineapple juice with MD (82%), but higher than those obtained by Kha et al.
268 (2010) (37-38%) and Ahmed et al. (2010) (42-57%) for the encapsulation of Gac fruit extract and
269 purple sweet potato, respectively. These differences can be attributed mainly to a higher content of
270 liposoluble substances in the raw material, which leads to high levels of insoluble components in the
271 extract (Kha et al., 2010) and consequently to a lower WSI.

272 At IT130, WSI seems to be not affected by MD concentration and feed flow ($p < 0.05$), similarly to
273 Kha et al. (2010), while Ahmed et al. (2010) stated that a lower WSI was related to a lower MD
274 concentration. Passing from IT130 to IT160, for the samples at MD10 a significant decrease in WSI
275 was noted (from $84 \pm 1.7\%$ to $79 \pm 2.2\%$ and from $85 \pm 2.2\%$ to $80 \pm 1.6\%$), while for the samples at
276 MD50 differences were not statistically significant ($p < 0.05$).

277 WAI were comprised between 0.28 and 0.58 g/g_{DP}, considerably lower than the results obtained by
278 Ahmed et al. (2010), which reported WAI comprised in the range 0.86-1.20 g/g_{DP}. The variation in
279 WAI can be related to different degrees of engagement of hydroxyl groups to form hydrogen and
280 covalent bonds between starch chains (Ahmed et al., 2010). Moreover, an increase in WAI has always
281 been associated with the loss of crystalline structure (Ahmed et al., 2010; Gunaratne & Hoover, 2002).
282 Increasing IT from IT130 to IT160, WAI significantly increased, suggesting a formation of mainly
283 amorphous MD structures during drying phase.

284 This behavior is also confirmed by the high SCs that characterize the powder products, considering
285 that the presence of a large number of crystallites, which increase granular stability, can be related to
286 low SC (Ahmed et al., 2010; Gunaratne & Hoover, 2002). For OPE microparticles, SC are comprised
287 between 2.10 ± 0.14 and 3.12 ± 0.14 g/g_{DP}, similar to those obtained by Ahmed et al. (2010), comprised

288 between 1.92 g/g_{DP} and 2.56 g/g_{DP}, and none of the parameters tested seems to have a significant
289 influence on SC. Only the sample spray dried at IT160, MD50 and feed flow 5 mL/min resulted
290 statistically different from other microparticles.

291

292 3.2.3 Thermogravimetric analysis

293 All the samples shown a weight loss between room temperature and 150 °C, due to powder moisture
294 (**Fig. 3a-h**). As reported in literature (Bilici, Doğan, Yıldırım, & Kaya, 2013; Dilek, Doğan, Bilici, &
295 Kaya, 2011), a more evident change in the TG curve begins at 185-220 °C until 500-550 °C, leading
296 to a global weight loss of 71-78%. Only samples spray dried at MD10 and feed flow of 5 mL/min,
297 for both IT tested, show a minor value, equal to 62 and 67% respectively. Residual ashes after the
298 combustion at 750 °C are included between 3 and 11%.

299

Fig.3

300 More interesting results were obtained by the analysis of differential thermogravimetric (DTG)
301 curves. In the temperature range of 200-300 °C, all the microparticles show a double peak. A first
302 endothermic reaction, shown in the differential thermal analysis curves (DTA), starts at 195 °C for
303 MD10 samples (**Fig. 3a-b**) and at 215 °C for MD50 samples (**Fig. 3c-d**). This difference is reasonable
304 assignable at the presence of MD, and it proves their thermal protection on the polyphenols. This
305 assumption is confirmed by the thermal analysis of polyphenol without MD, where already at IT160
306 is evident a weight loss due to decomposition (**Fig. 3j**). At a temperature of about 250 °C, a second
307 reaction starts, with a maximum rate at 280-290 °C, depending on the MD amount. This second
308 reaction is predominant when higher is the MD concentration as well as when higher is the feed flow,
309 and it can be ascribed at the MD decomposition (**Fig. 3i**).

310 All the DTA curves show only an endothermic broad peak connected with the melting point of the
311 microparticles (around 230 °C), while to the weight loss do not correspond any significant change,
312 probably due to the presence of exothermic phenomena that offset the endothermic pyrolysis reaction.

313 No significant differences in the thermal behavior between the samples dried at IT130 and IT160
314 were evident.

315

316 *3.3 TP, ARP, microencapsulation yield and efficiency*

317 **Table 3** shows TP content, ARP, microencapsulation yield and efficiency of the microparticles.

318

Table 3

319 TP content is maximum (39.5 ± 4.9 mg_{CAE/g_DFP}) at IT130, MD10 and feed flow of 10 mL/min. Except
320 for this case, TP in the products is not influenced by IT, emphasizing the thermo-protective effect of
321 MD (Munin & Edwards-Lévy, 2011). This effect is also evident considering ARP values of
322 microparticles: passing from IT130 to IT160, for the samples spray dried with MD10 a significant
323 decrease ($p < 0.05$) of ARP was noted, while in samples at MD50 this phenomenon is not present.

324 The highest ARPs were found in microparticles with MD10 and feed flow equal to 10 mL/min, for
325 both the IT tested (33.8 ± 4.3 and 19.1 ± 2.1 mmol_{DPPH}/L_{extract} for IT130 and IT160, respectively).

326 Microencapsulation yields are higher than 50% for all samples, and the maximum yield was obtained
327 at IT130, MD10 and feed flow 10 mL/min. At these conditions, $94 \pm 0.4\%$ of the polyphenols of OPE
328 is still present in powders, suggesting that polyphenols were not affected by high inlet temperatures
329 used in the drying process.

330 This was confirmed by HPLC-DAD, in which only small differences could be noted in the phenolic
331 profiles of microparticles (**Table 4**). Indeed, an increase in IT leads to a significant difference in the
332 phenolic profiles only between powders obtained at MD10 and feed flow 5 mL/min (from 27.9 ± 1.1
333 to $21.1 \pm 1.2\%$ and from 12.1 ± 1.0 to $18.5 \pm 1.8\%$ for tyrosol and apigenin 7-glucoside, respectively)
334 and between samples obtained at MD10 and feed flow of 10 mL/min for caffeic acid (from 9.6 ± 1.0
335 to $4.4 \pm 0.8\%$).

336

Table 4

337 Compared to OPE, before encapsulation microparticles show a different profile of polyphenols. This
338 is evident for tyrosol, luteolin 7-glucoside and luteolin, in which their percentages, compared to the

339 total identified polyphenols, are higher in OPE (equal to 35 ± 1.2 , 27 ± 0.5 and $8\pm 0.4\%$, respectively)
340 than in microparticles, indicating that these compounds are more sensitive to spray drying than other
341 polyphenols.

342 ME resulted higher than 65% for all samples, and at IT160 MEs were similar or lower to the ones
343 obtained at IT130: this can be explained considering that the evaporation rate is higher when IT
344 increases, generating a lower bulk density, as shown in **Table 1**, and consequently a higher
345 microparticles surface exposed to the environment. ME seems to be not influenced by feed flow. The
346 higher ME was obtained at IT130, MD10 and feed flow at 5 mL/min, equal to $77\pm 3.5\%$.

347

348 *3.4 Stability tests*

349 The best sample in terms of microencapsulation yield and ME was selected for stability tests at three
350 different temperatures and three different light conditions for 70 days (**Fig. 4a-b**). Storage under UV
351 light conditions was also studied for a period of 48 h (**Fig. 4c**). TP and ARP were measured.

352

Fig.4

353 The role of storage temperature in the degradation of encapsulated polyphenols at 5, 25 and 45 °C is
354 shown in **Fig. 4a**. At 25 °C, microparticles were stable until 28 days of storage ($p < 0.05$), while at
355 45 °C a statistically significant decrease was found already after 14 days of storage. During the
356 storage, microparticles stored at 5 °C had shown no significant differences compared to the initial
357 sample, suggesting that this condition can be ideal for a long storage period of the powders. After 70
358 days, a reduction of 21% and 34% of the initial TP was noted for the samples stored at 25 °C and 45
359 °C, respectively.

360 **Fig. 4b** shows the effect of different light conditions on TP of the microparticles, at room temperature.
361 Powders stored in dark conditions show a limited degradation during storage, probably due to the
362 temperature, as described above. Samples exposed to sunlight had shown a statistically significant
363 difference after 14 days, decreasing from 40.0 ± 4.0 mg_{CAE}/g_{DP} to 33.1 ± 0.8 mg_{CAE}/g_{DP}. After 70 days,
364 TP of the microparticles was decreased up to 27.7 ± 0.7 mg_{CAE}/g_{DP}, indicated that 31% of the initial

365 polyphenols have been oxidized. At every time considered, TP contents were similar to those obtained
366 in dark conditions ($p < 0.05$), suggesting that the effect of temperature on the stability of the product
367 is more evident than that of sunlight. Under strong light illuminance (2.1 klux), the degradation of
368 polyphenols is accelerated: after 7 days of storage a reduction in TP content was detected ($p < 0.05$),
369 and after 70 days 44% of the initial polyphenols were degraded. Considering these results, a
370 temperature equal to 5 °C in dark conditions seem to be the best choice for a long-period storage of
371 the spray dried OP polyphenols.

372 The effect of UV light on microparticles was evaluated during 48 h storage under UV lamp (**Fig. 4c**).
373 After 1 h, 33% of the initial polyphenols were degraded, probably due to oxidation of surface
374 polyphenols, which for that sample are equal to $9.4 \pm 0.4 \text{ mg}_{\text{CAE}}/\text{g}_{\text{DP}}$. In the storage period from 1 h to
375 8 h, microparticles had not shown TP differences ($p < 0.05$), while after 8 h TP decreased up to
376 $13.7 \pm 0.8 \text{ mg}_{\text{CAE}}/\text{g}_{\text{DP}}$ after 48 h of exposure, with the loss of 66% of initial TP content.
377 For all the storage conditions tested, ARP values follow the same behaviors of TP values.

378

379 **4. Conclusions**

380 Results of this study confirmed the efficiency of spray drying to microencapsulate OP polyphenols.
381 The effect of IT, MD concentration and feed flow were studied and microparticles were characterized.
382 An increase from IT130 to IT160 leads to lower moisture content and bulk density, but has not
383 statistically significant ($p < 0.05$) effects on microencapsulation yield and ME. Increasing MD
384 concentration causes lower bulk density and higher size of the powders. WAI and SC values suggests
385 that by increasing IT, the structure of the MD matrix tends to be in an amorphous state, which enhance
386 the final solubility of the samples. The highest microencapsulation yield was obtained at IT130,
387 MD10 and feed flow of 10 mL/min. At these conditions notable TP content ($39.5 \pm 4.9 \text{ mg}_{\text{CAE}}/\text{g}_{\text{DP}}$)
388 and ARP ($33.8 \pm 4.3 \text{ mmol}_{\text{DPPH}}/\text{L}_{\text{extract}}$) were obtained, along with a high ME ($76 \pm 3.3\%$), and small
389 particle size.

390 The product obtained shows good stability at storage conditions and remarkable antioxidant
391 properties, and can be considered a potential source for integration in novel food or pharmaceutical
392 products.

393

394 **References**

395 Abadio, F. D. B., Domingues, A. M., Borges, S., & Oliveira, V. M. (2004). Physical properties of
396 powdered pineapple (*Ananas comosus*) juice-effect of malt dextrin concentration and
397 atomization speed. *Journal of Food Engineering*, *64*, 285-287.

398 Ahmed, M., Akter, M. S., Lee, J. C., & Eun, J. B. (2010). Encapsulation by spray drying of bioactive
399 components, physicochemical and morphological properties from purple sweet potato. *LWT-
400 Food Science and Technology*, *43*, 1307-1312.

401 Aliakbarian, B., Casazza, A. A., & Perego, P. (2011). Valorization of olive oil solid waste using high
402 pressure-high temperature reactor. *Food Chemistry*, *128*, 704-710.

403 Aliakbarian, B., Palmieri, D., Casazza, A. A., Palombo, D., & Perego, P. (2012). Antioxidant activity
404 and biological evaluation of olive pomace extract. *Natural Product Research*, *26*, 2280-2290.

405 An, H. J., Wilhelm, W. E., & Searcy, S. W. (2011). Biofuel and petroleum-based fuel supply chain
406 research: A literature review. *Biomass & Bioenergy*, *35*, 3763-3774.

407 Bilici, A., Doğan, F., Yıldırım, M., & Kaya, İ. (2013). Facile synthesis of self-stabilized polyphenol
408 nanoparticles. *Materials Chemistry and Physics*, *140*, 66-74.

409 Casazza, A. A., Aliakbarian, B., De Faveri, D., Fiori, L., & Perego, P. (2012). Antioxidants from
410 winemaking wastes: a study on extraction parameters using response surface methodology.
411 *Journal of Food Biochemistry*, *36*, 28-37.

412 Champagne, P. (2007). Feasibility of producing bio-ethanol from waste residues: A Canadian
413 perspective Feasibility of producing bio-ethanol from waste residues in Canada. *Resources
414 Conservation and Recycling*, *50*, 211-230.

415 Che, F., Sarantopoulos, I., Tsoutsos, T., & Gekas, V. (2012). Exploring a promising feedstock for
416 biodiesel production in Mediterranean countries: A study on free fatty acid esterification of
417 olive pomace oil. *Biomass & Bioenergy*, *36*, 427-431.

418 Cioffi, G., Pesca, M. S., De Caprariis, P., Braca, A., Severino, L., & De Tommasi, N. (2010). Phenolic
419 compounds in olive oil and olive pomace from Cilento (Campania, Italy) and their antioxidant
420 activity. *Food Chemistry*, *121*, 105-111.

421 da Costa, E. M., Barbosa, J. M., do Nascimento, T. G., & Macedo, R. O. (2002). Thermal
422 characterization of the quercetin and rutin flavonoids. *Thermochimica Acta*, *392*, 79-84.

423 de Vos, P., Faas, M. M., Spasojevic, M., & Sikkema, J. (2010). Encapsulation for preservation of
424 functionality and targeted delivery of bioactive food components. *International Dairy*
425 *Journal*, *20*, 292-302.

426 Desai, K. G. H., & Park, H. J. (2005). Recent developments in microencapsulation of food
427 ingredients. *Drying Technology*, *23*, 1361-1394.

428 Dilek, D., Doğan, F., Bilici, A., & Kaya, İ. (2011). Oxidative synthesis of a novel polyphenol having
429 pendant Schiff base group: Synthesis, characterization, non-isothermal decomposition
430 kinetics. *Thermochimica Acta*, *518*, 72-81.

431 Ersus, S., & Yurdagel, U. (2007). Microencapsulation of anthocyanin pigments of black carrot
432 (*Daucuscarota* L.) by spray drier. *Journal of Food Engineering*, *80*, 805-812.

433 Fang, Z. X., & Bhandari, B. (2012). Comparing the efficiency of protein and maltodextrin on spray
434 drying of bayberry juice. *Food Research International*, *48*, 478-483.

435 Fazaeli, M., Emam-Djomeh, Z., Ashtari, A. K., & Omid, M. (2012). Effect of spray drying conditions
436 and feed composition on the physical properties of black mulberry juice powder. *Food and*
437 *Bioproducts Processing*, *90*, 667-675.

438 Gharsallaoui, A., Roudaut, G., Chambin, O., Voilley, A., & Saurel, R. (2007). Applications of spray-
439 drying in microencapsulation of food ingredients: An overview. *Food Research International*,
440 *40*, 1107-1121.

- 441 Gunaratne, A., & Hoover, R. (2002). Effect of heat–moisture treatment on the structure and
442 physicochemical properties of tuber and root starches. *Carbohydrate Polymers*, *49*, 425-437.
- 443 Haddadin, M. S. Y., Haddadin, J., Arabiyat, O. I., & Hattar, B. (2009). Biological conversion of olive
444 pomace into compost by using *Trichoderma harzianum* and *Phanerochaete chrysosporium*.
445 *Bioresource Technology*, *100*, 4773-4782.
- 446 Kha, T. C., Nguyen, M. H., & Roach, P. D. (2010). Effects of spray drying conditions on the
447 physicochemical and antioxidant properties of the Gac (*Momordica cochinchinensis*) fruit aril
448 powder. *Journal of Food Engineering*, *98*, 385-392.
- 449 Lai, H. M., & Cheng, H. H. (2004). Properties of pregelatinized rice flour made by hot air or gum
450 puffing. *International Journal of Food Science and Technology*, *39*, 201-212.
- 451 Makris, D. P., Boskou, G., & Andrikopoulos, N. K. (2007). Polyphenolic content and in vitro
452 antioxidant characteristics of wine industry and other agri-food solid waste extracts. *Journal*
453 *of Food Composition and Analysis*, *20*, 125-132.
- 454 Mavros, M., Xekoukoulotakis, N. P., Mantzavinos, D., & Diamadopoulos, E. (2008). Complete
455 treatment of olive pomace leachate by coagulation, activated-carbon adsorption and
456 electrochemical oxidation. *Water Research*, *42*, 2883-2888.
- 457 Michailides, M., Christou, G., Akrotos, C. S., Tekerlekopoulou, A. G., & Vayenas, D. V. (2011).
458 Composting of olive leaves and pomace from a three-phase olive mill plant. *International*
459 *Biodeterioration & Biodegradation*, *65*, 560-564.
- 460 Miranda, T., Arranz, J. I., Montero, I., Roman, S., Rojas, C. V., & Nogales, S. (2012).
461 Characterization and combustion of olive pomace and forest residue pellets. *Fuel Processing*
462 *Technology*, *103*, 91-96.
- 463 Munin, A., & Edwards-Lévy, F. (2011). Encapsulation of Natural Polyphenolic Compounds; a
464 Review. *Pharmaceutics*, *3*, 793-829.

465 Pagnanelli, F., Mainelli, S., Veglio, F., & Toro, L. (2003). Heavy metal removal by olive pomace:
466 biosorbent characterisation and equilibrium modelling. *Chemical Engineering Science*, *58*,
467 4709-4717.

468 Palmieri, D., Aliakbarian, B., Casazza, A. A., Ferrari, N., Spinella, G., Pane, B., Cafueri, G., Perego,
469 P., & Palombo, D. (2012). Effects of polyphenol extract from olive pomace on anoxia-induced
470 endothelial dysfunction. *Microvascular Research*, *83*, 281-289.

471 Robert, P., Gorena, T., Romero, N., Sepulveda, E., Chavez, J., & Saenz, C. (2010). Encapsulation of
472 polyphenols and anthocyanins from pomegranate (*Punica granatum*) by spray drying.
473 *International Journal of Food Science and Technology*, *45*, 1386-1394.

474 Suarez, M., Romero, M. P., Ramo, T., Macia, A., & Motilva, M. J. (2009). Methods for preparing
475 phenolic extracts from olive cake for potential application as food antioxidants. *Journal of*
476 *Agricultural and Food Chemistry*, *57*, 1463-1472.

477 Tan, L. W., Ibrahim, M. N., Kamil, R., & Taip, F. S. (2011). Empirical modeling for spray drying
478 process of sticky and non-sticky products. *Procedia Food Science*, *1*, 690-697.

479 Tekin, A. R., & Dalgic, A. C. (2000). Biogas production from olive pomace. *Resources Conservation*
480 *and Recycling*, *30*, 301-313.

481 Tuck, K. L., & Hayball, P. J. (2002). Major phenolic compounds in olive oil: metabolism and health
482 effects. *Journal of Nutritional Biochemistry*, *13*, 636-644.

483

484 **Figure Captions**

485

486 **Fig.1** SEM images of the microcapsules obtained at different inlet temperature (IT), maltodextrins
487 (MD) concentrations and feed flows: a) IT 130°C, MD 100 g/L, feed flow 10 mL/min; b) IT 130°C,
488 MD 500 g/L, feed flow 10 mL/min; c) IT 160°C, MD 100 g/L, feed flow 10 mL/min; d) IT 160°C,
489 MD 500 g/L, feed flow 5 mL/min

490

491 **Fig.2** Percentage of microparticles with diameter comprised in different size ranges, calculated as the
492 ratio between the number of particles in the range and the number of total measured particles. a)

493 Microparticles obtained at IT130, MD10, feed flow 5 mL/min (■) and IT130, MD10, feed flow 10
494 mL/min (□). b) Microparticles obtained at IT130, MD50, feed flow 5 mL/min (■) and IT130, MD50,
495 feed flow 10 mL/min (□). c) Microparticles obtained at IT160, MD10, feed flow 5 mL/min (■) and
496 IT160, MD10, feed flow 10 mL/min (□). d) Microparticles obtained at IT160, MD50, feed flow 5
497 mL/min (■) and IT160, MD50, feed flow 10 mL/min (□).

498

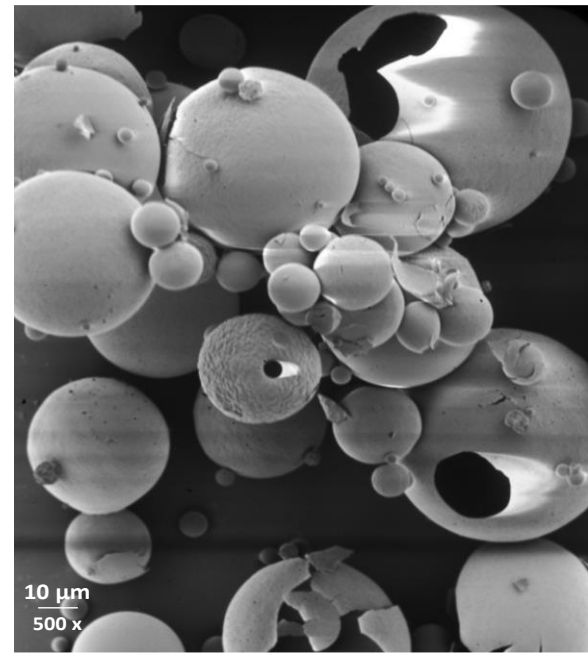
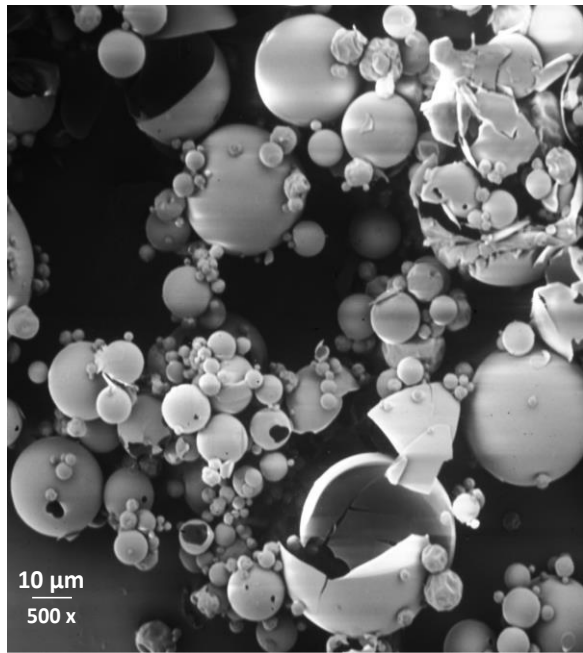
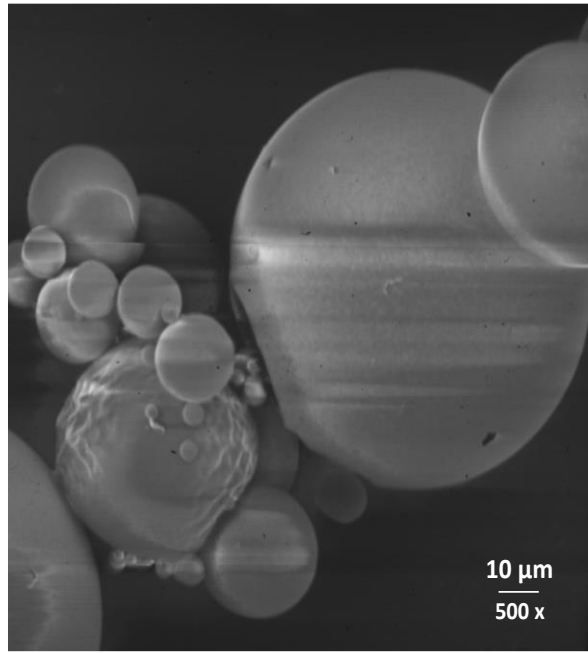
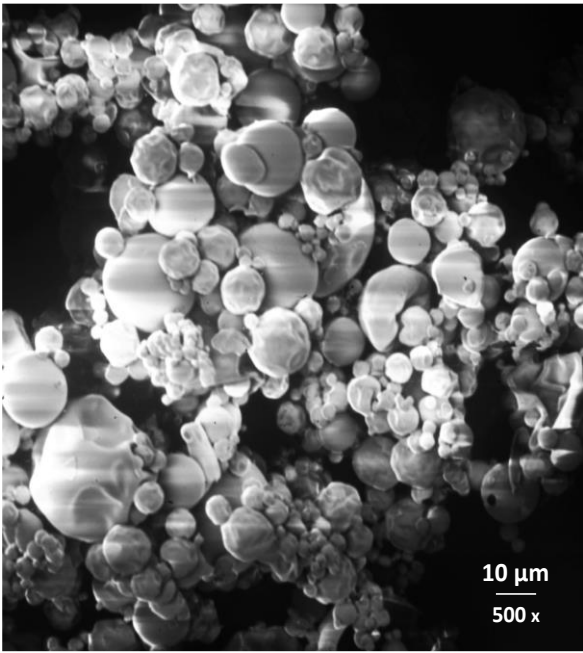
499 **Fig.3** Thermogravimetric analysis (— TG), differential thermogravimetric analysis (... DTG) and
500 differential thermal analysis (— DTA) of the microparticles obtained at different conditions: a)
501 IT130, MD10, feed flow 5 mL/min; b) IT130, MD10, feed flow 10 mL/min; c) IT130, MD50, feed
502 flow 5 mL/min; d) IT130, MD50, feed flow 10 mL/min, e) IT160, MD10, feed flow 5 mL/min; f)
503 IT160, MD10, feed flow 10 mL/min; g) IT160, MD50, feed flow 5 mL/min; h) IT160, MD50, feed
504 flow 10 mL/min; i) MD without extract at IT130, MD10, feed flow 10 mL/min; j) Extract without
505 MD.

506

507 **Fig.4** Effect of different storage conditions (temperature, light, UV exposure) on microparticles. a)
508 TP of stored microparticles at 5 °C (—), 25 °C (— —) and 45 °C (...) during a 10 weeks storage.
509 b) TP of stored microparticles in dark conditions (—), sunlight (— —) and lamp light (...) for a 10
510 weeks storage. c) TP of stored microparticles under UV exposure during a 48 h storage. Different
511 letters indicate significant difference between data at $p < 0.05$.

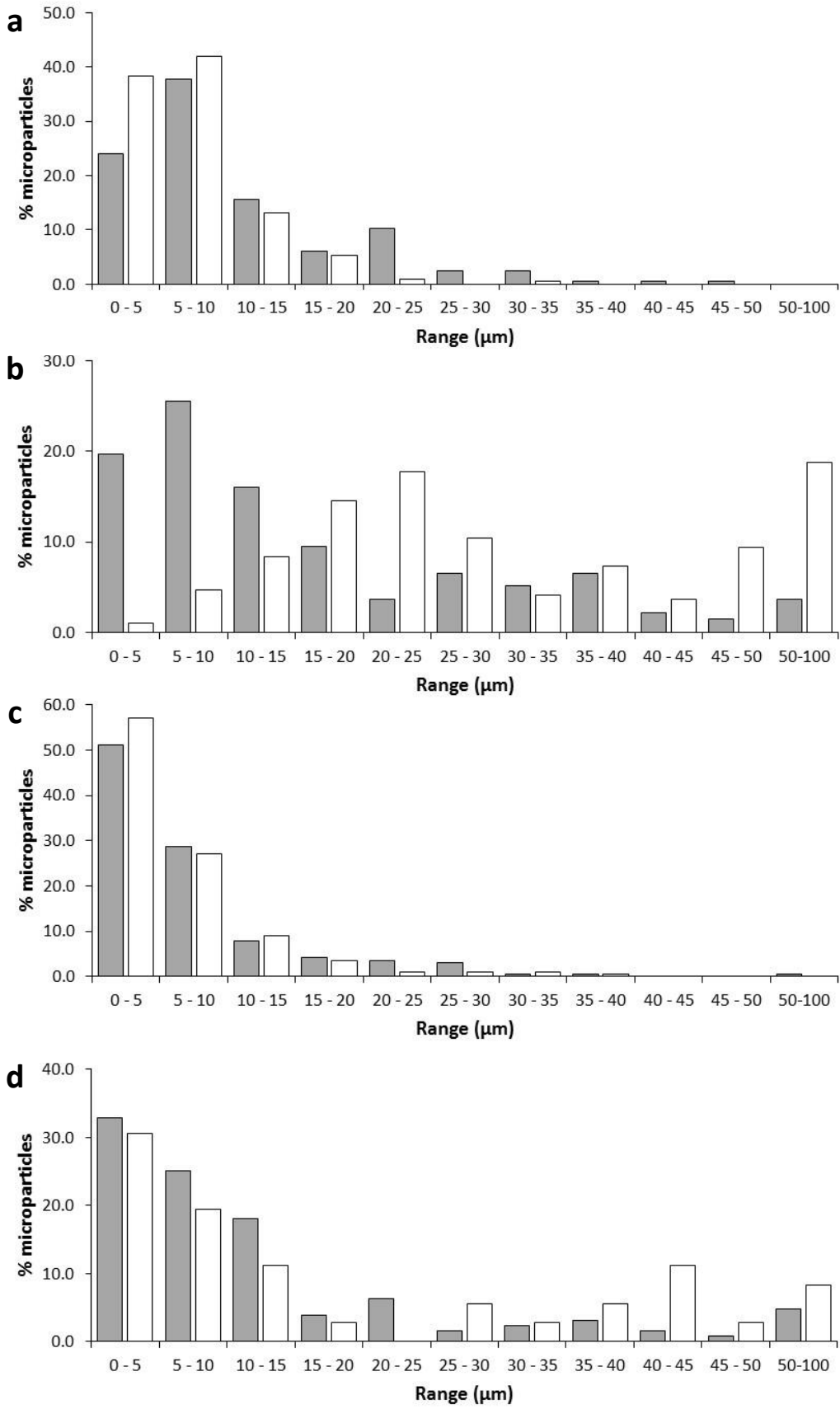
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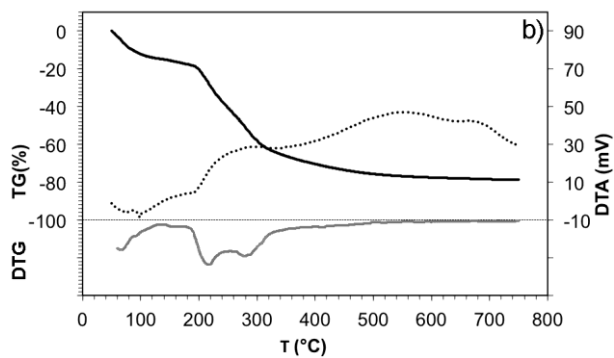
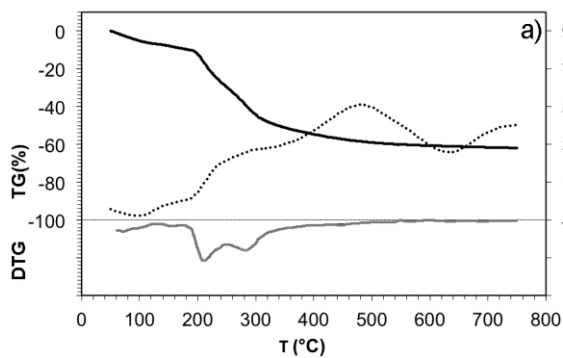
Fig.1



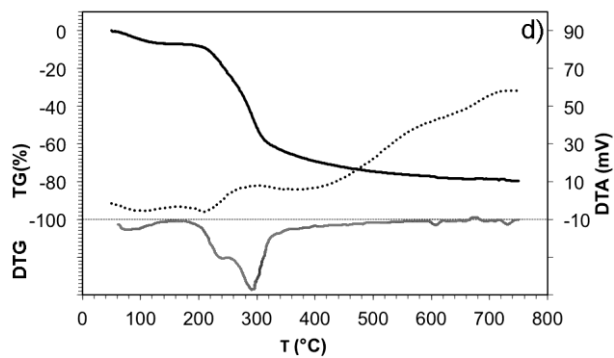
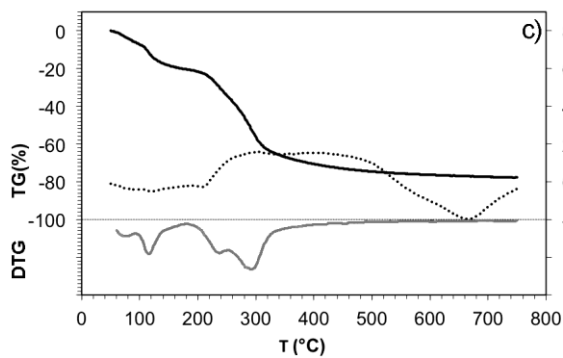
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Fig. 2

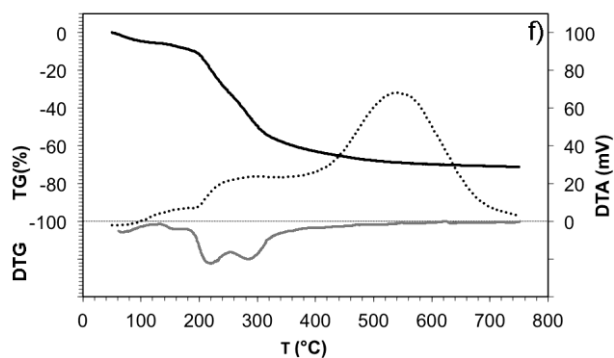
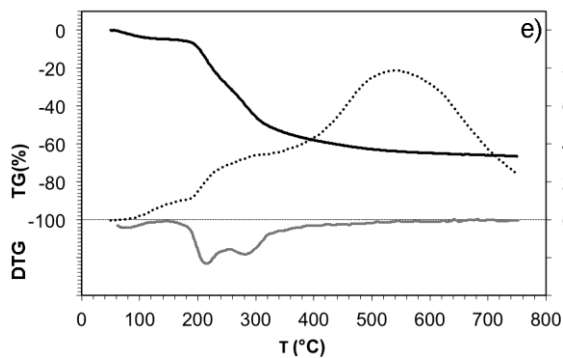
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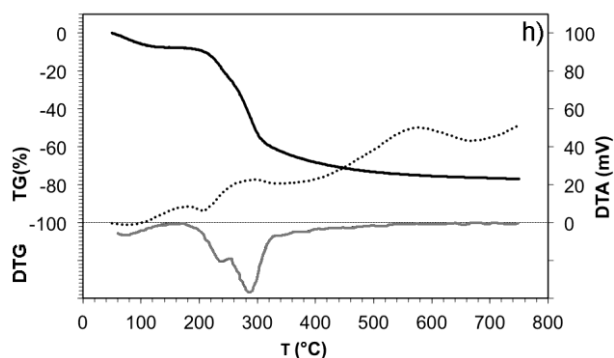
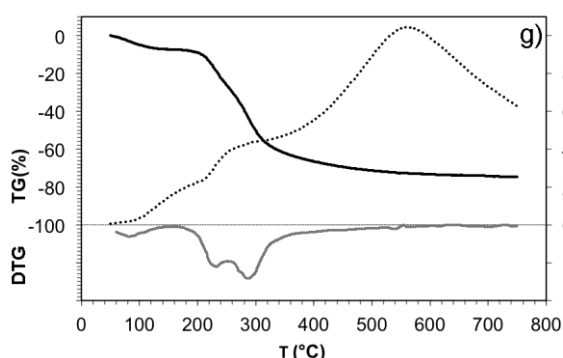
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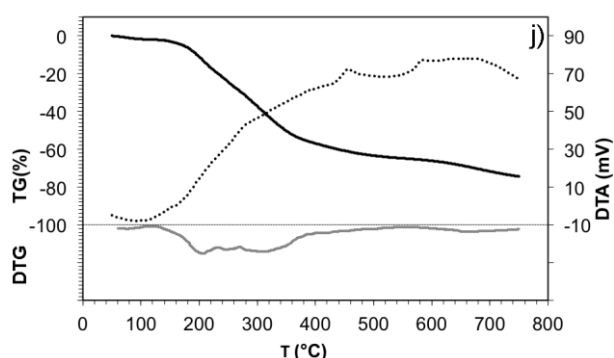
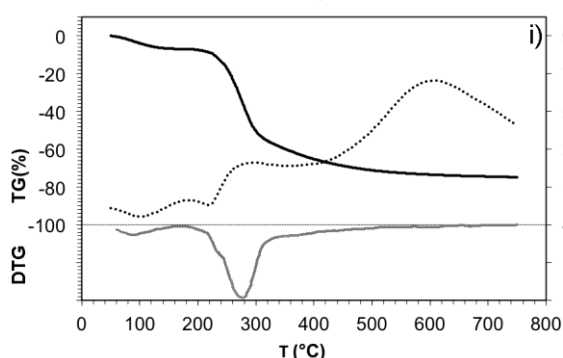
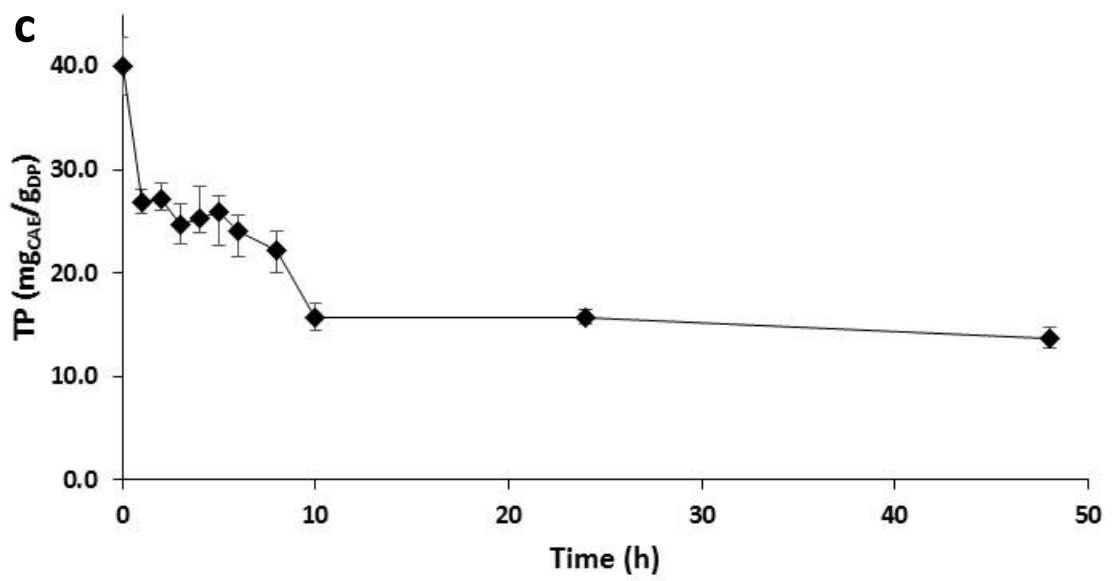
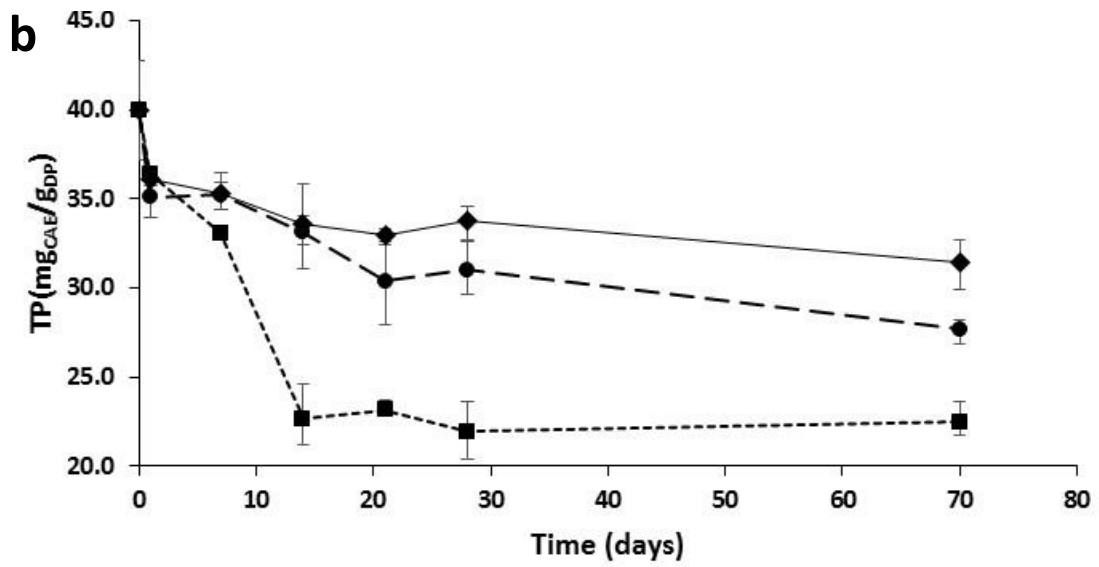
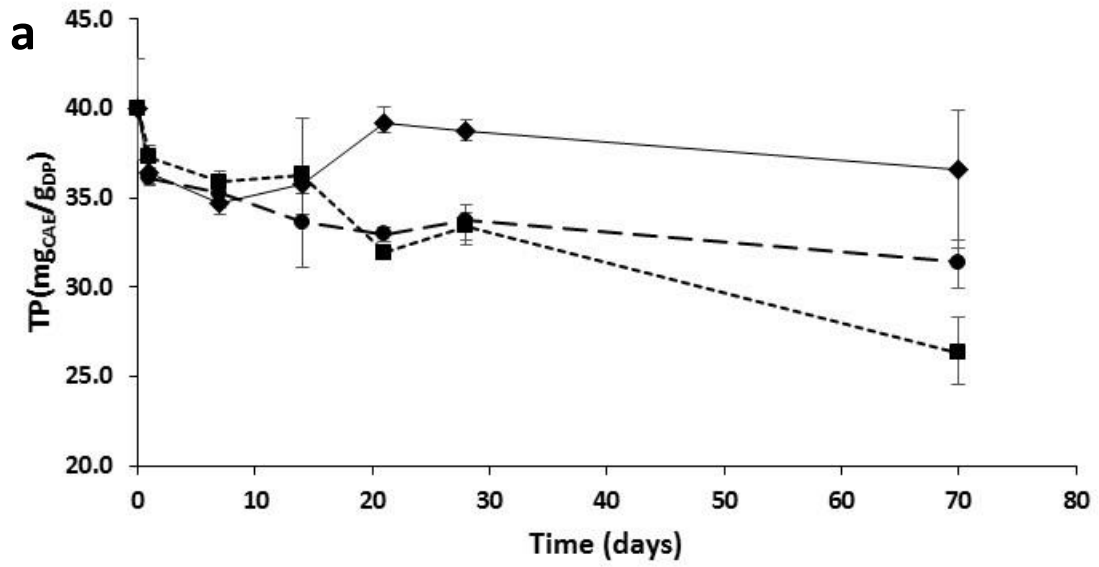


Fig.3



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528
529

Fig.4

530 **Table 1**
 531 Physical characterization of the microcapsules obtained by SD of OP extract using MD as coating
 532 agent.

IT (°C)	MD (g/L)	Feed flow (mL/min)	OT (°C)	Moisture content (%)	Bulk density (g/mL)
130	100	5	86 ± 1 ^{a,b}	6.2 ± 0.2 ^b	0.19 ± 0.01 ^b
		10	75 ± 1 ^c	7.0 ± 0.4 ^c	0.19 ± 0.02 ^b
	500	5	87 ± 1 ^{a,b}	5.9 ± 0.4 ^b	0.13 ± 0.01 ^e
		10	83 ± 1 ^a	6.1 ± 0.2 ^{b,c}	0.16 ± 0.01 ^f
160	100	5	95 ± 1 ^d	2.3 ± 0.3 ^a	0.07 ± 0.01 ^c
		10	84 ± 1 ^a	2.1 ± 0.5 ^a	0.10 ± 0.01 ^d
	500	5	102 ± 1 ^e	2.4 ± 0.3 ^a	0.04 ± 0.01 ^a
		10	89 ± 1 ^b	2.4 ± 0.4 ^a	0.05 ± 0.01 ^a

533 Different letters within the column indicate significant difference between data at $p < 0.05$. Data are
 534 expressed as mean value ± standard deviation, $n = 3$. For bulk density measurements, $n = 5$. (SD:
 535 Spray Drying, OP: Olive Pomace, MD: Maltodextrins, IT: Inlet Temperature, OT: Outlet
 536 Temperature).
 537

538 **Table 2**
 539 WSI, WAI and SC of the microparticles obtained by SD of OP extract using MD as coating agent.

IT (°C)	MD (g/L)	Feed flow (mL/min)	WSI (%)	WAI (g/gDP)	SC (g/gDP)
130	100	5	84 ± 1.7 ^a	0.34 ± 0.02 ^a	2.13 ± 0.27 ^a
		10	85 ± 2.2 ^a	0.33 ± 0.02 ^{a,b}	2.31 ± 0.49 ^a
	500	5	85 ± 0.9 ^a	0.38 ± 0.01 ^{a,c}	2.49 ± 0.18 ^{a,b}
		10	87 ± 1.6 ^a	0.28 ± 0.02 ^b	2.10 ± 0.14 ^a
160	100	5	79 ± 2.2 ^b	0.49 ± 0.02 ^d	2.43 ± 0.33 ^{a,b}
		10	80 ± 1.6 ^b	0.58 ± 0.03 ^e	2.85 ± 0.16 ^{a,b}
	500	5	83 ± 0.5 ^{a,b}	0.53 ± 0.01 ^d	3.12 ± 0.14 ^b
		10	85 ± 1.2 ^a	0.40 ± 0.02 ^c	2.76 ± 0.31 ^{a,b}

540 Different letters within the column indicate significant difference between data at $p < 0.05$. Data are
 541 expressed as mean value ± standard deviation, $n = 3$ (WSI: Water Solubility Index, WAI: Water
 542 Absorption Index, SC: Swelling Capacity, SD: Spray Drying, OP: Olive Pomace, MD: Maltodextrins,
 543 DP: Dried Powder).
 544

545 **Table 3**
 546 TP, ARP, encapsulation yield and ME of the microparticles obtained by SD of OP extract using MD
 547 as coating agent.

IT (°C)	MD (g/L)	Feed flow (mL/min)	TP (mg _{CAE} /g _{DPE})	ARP (mmol _{DPPH} /L _{extract})	Microencapsulation Yield (%)	ME (%)
130	100	5	27.7 ± 1.7 ^b	20.4 ± 2.7 ^c	66 ± 4.1 ^{a, b}	77 ± 3.5 ^b
		10	39.5 ± 4.9 ^c	33.8 ± 4.3 ^d	94 ± 0.4 ^c	76 ± 3.3 ^{a, b}
	500	5	4.5 ± 0.3 ^a	3.8 ± 0.6 ^a	54 ± 4.1 ^a	72 ± 1.4 ^{a, b}
		10	6.2 ± 0.4 ^a	6.8 ± 1.0 ^a	73 ± 0.6 ^b	75 ± 1.0 ^{a, b}
160	100	5	26.0 ± 1.2 ^b	14.8 ± 0.7 ^b	62 ± 2.9 ^{a, b}	65 ± 3.5 ^a
		10	26.0 ± 1.5 ^b	19.1 ± 2.1 ^c	60 ± 3.6 ^{a, b}	66 ± 3.1 ^{a, b}
	500	5	4.8 ± 0.5 ^a	5.0 ± 0.4 ^a	58 ± 7.1 ^{a, b}	73 ± 5.2 ^{a, b}
		10	4.4 ± 0.3 ^a	3.6 ± 0.3 ^a	51 ± 4.8 ^a	71 ± 6.8 ^{a, b}

548 Different letters within the column indicate significant difference between data at $p < 0.05$. Data are
 549 expressed as mean value ± standard deviation, $n = 3$ (TP: Total polyphenols, ARP: Antiradical Power,
 550 ME: Microencapsulation Efficiency, SD: Spray Drying, OP: Olive Pomace, MD: Maltodextrins, IT:
 551 Inlet Temperature).

552

553

Table 4

554

Percentage of each phenolic compound respect to the total amount of identified polyphenols in the olive pomace (OP) extract before spray drying and in microparticles spray dried at different inlet temperatures (IT), maltodextrins concentration (MD) and feed flows.

555

IT (°C)	MD (g/L)	Feed flow (mL/min)	¹ Tyr	² VA	³ SA	⁴ CA	⁵ <i>p</i> -CA	⁶ Lut 7-G	⁷ Api 7-G	⁸ Lut	⁹ Api
130	100	5	27.9 ± 1.1 ^b	16.5 ± 1.3 ^a	3.6 ± 0.6 ^a	6.3 ± 1.0 ^b	3.2 ± 0.4 ^a	10.0 ± 1.2 ^a	12.1 ± 1.0 ^b	5.9 ± 0.5 ^a	14.6 ± 1.3 ^c
		10	19.5 ± 3.9 ^a	16.5 ± 0.9 ^a	4.7 ± 0.5 ^a	9.6 ± 1.0 ^c	1.9 ± 0.7 ^a	13.6 ± 2.1 ^a	14.9 ± 1.8 ^{b,c}	6.0 ± 0.4 ^a	13.3 ± 1.0 ^c
	500	5	21.9 ± 1.0 ^a	14.1 ± 0.5 ^a	4.0 ± 0.4 ^a	6.0 ± 1.2 ^b	3.0 ± 1.0 ^a	22.7 ± 1.3 ^b	10.7 ± 1.0 ^b	6.3 ± 0.4 ^a	11.1 ± 0.6 ^b
		10	24.6 ± 1.8 ^a	18.5 ± 1.0 ^a	4.9 ± 0.6 ^a	4.9 ± 1.4 ^b	3.6 ± 1.0 ^a	15.0 ± 1.9 ^a	13.7 ± 1.1 ^b	4.4 ± 0.8 ^a	10.4 ± 0.5 ^b
160	100	5	21.1 ± 1.2 ^a	14.8 ± 1.8 ^a	3.3 ± 1.0 ^a	6.8 ± 1.5 ^b	3.5 ± 0.3 ^a	12.2 ± 1.0 ^a	18.5 ± 1.8 ^c	5.5 ± 0.4 ^a	14.2 ± 1.0 ^c
		10	18.4 ± 3.9 ^a	18.6 ± 1.4 ^a	5.4 ± 1.4 ^a	4.4 ± 0.8 ^b	1.7 ± 0.9 ^a	13.7 ± 0.9 ^a	17.3 ± 2.0 ^c	6.5 ± 0.6 ^a	14.0 ± 0.9 ^c
	500	5	23.9 ± 0.8 ^a	14.3 ± 1.9 ^a	3.8 ± 1.2 ^a	4.4 ± 1.0 ^b	2.9 ± 0.5 ^a	22.2 ± 1.0 ^b	12.9 ± 1.5 ^b	5.4 ± 0.7 ^a	10.2 ± 0.7 ^b
		10	25.2 ± 1.0 ^a	14.3 ± 1.6 ^a	4.1 ± 1.1 ^a	4.4 ± 1.1 ^b	3.3 ± 0.6 ^a	19.4 ± 2.5 ^{a,b}	13.4 ± 0.8 ^b	5.7 ± 0.5 ^a	10.4 ± 0.4 ^b
OP extract			35.1 ± 1.2 ^c	12.9 ± 1.3 ^a	2.9 ± 0.8 ^a	2.9 ± 0.5 ^a	1.1 ± 0.8 ^a	26.5 ± 0.5 ^c	4.5 ± 0.8 ^a	8.1 ± 0.4 ^b	5.9 ± 0.6 ^a

556

Different letters within the column indicate significant difference between data at $p < 0.05$. Data are expressed as mean value ± standard deviation, $n = 3$.

557

558

¹Tyr: Tyrosol.

559

²VA: Vanillic acid.

560

³SA: Syringic acid.

561

⁴CA: Caffeic acid.

562

⁵*p*-CA: *p*-coumaric acid.

563

⁶Lut 7-G: Luteolin 7-glucoside.

564

⁷Api 7-G: Apigenin 7-glucoside.

565

⁸Lut: Luteolin.

566

⁹Api: Apigenin.

567