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# Insights into the Pharmacological Activity of the Imidazo—Pyrazole Scaffold

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In previous studies, we synthesized different imidazo—pyrazoles 1 and 2 with interesting anticancer, anti-angiogenic and anti-inflammatory activities. To further extend the structure-activity relationships of imidazo—pyrazole scaffold and to identify novel antiproliferative/anti-inflammatory agents potentially active with multi-target mechanisms, a library of compounds 3–5 has been designed and synthesized. The chemical modifications characterizing the novel derivatives include: i) decoration of the catechol ring with groups with different electronic, steric and lipophilic properties (compounds 3); ii) insertion of a methyl group on C-6 of imidazo—pyrazole scaffold (compounds 4); iii) shift of the acylhydrazonic substituent from position 7 to 6 of

the imidazo—pyrazole substructure (compounds 5). All synthesized compounds were tested against a panel of cancer and normal cell lines. Derivatives  $\bf 3a$ ,  $\bf 3e$ ,  $\bf 4c$ ,  $\bf 5g$  and  $\bf 5h$  showed IC<sub>50</sub> values in the low micromolar range against selected tumor cell lines and proved to have antioxidant properties, being able to inhibit ROS production in human platelet. In silico calculation predicted favourable drug-like and pharmacokinetic properties for the most promising compounds. Furthermore, molecular docking and molecular dynamic simulations suggested the ability of most active derivative  $\bf 3e$  to interact with colchicine binding site in the polymeric tubulin  $\alpha$ /tubulin  $\beta$ /stathmin4 complex.

### Introduction

Cancer is a multifactorial disease characterized by unchecked cell proliferation that quickly infiltrate in tissues and organs. Although many innovative therapies have been introduced to date, the onset of drug resistance is expected to cause a large increase in cancer mortality in the next few years.<sup>[1]</sup>

Although inflammation and cancer are often associated, the involvement of phlogosis and immunosurveillance in cancer development and progression has been established only in recent decades. [2,3] Indeed, it has been evaluated that about 20% of cancers are related to inflammation and furthermore solid neoplasms are preceded by tissue inflammation. [4-6] Moreover, different studies demonstrated that neutrophil migration (chemotaxis) and subsequent production of cytokines, most involved in inflammation process, play also a central role in

cancer progression and onset.<sup>[2,5]</sup> Collectively all these data support a close relationship between inflammation and cancer.

Differently from other heterocyclic compounds,<sup>[7,8]</sup> imidazo—pyrazole scaffold represents a chemotype not deeply investigated in medicinal chemistry and could provide important hits for the identification of novel pharmacologically active compounds.

In our previous studies, we identified a large library of imidazo—pyrazoles differently decorated on N1, C-6 and C-7 position (compounds 1, Figure 1) able to contrast inflammation by blocking neutrophils chemotaxis and inhibiting ERK1/2, AKT and p38MAPK phosphorylation.<sup>[9,10]</sup> Moreover, some derivatives proved to reduced angiogenesis process in VEGF-stimulated human umbilical vein endothelial cells (HUVEC) and in neuroblastoma cell lines.<sup>[11,12]</sup> More recently, other imidazo—pyrazoles (compounds 2, Figure 1), sharing with a number of new

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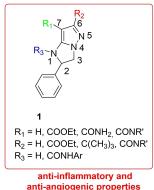




Figure 1. Structure of previous synthesized imidazo—pyrazoles 1 and 2 and related biological properties.

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anticancer drugs a differently decorated acylhydrazone moiety,[13-15] evidenced interesting antiproliferative and antioxidant properties.[16,17] Interestingly, compounds with bulky substituents at para position of catechol ring (derivatives 2a and 2b, Figure 2) evidenced a remarkable anticancer activity through the interaction with different intracellular systems, including tubulin,[16] whereas derivatives characterized by a smaller substituent (e.g., 2c, Figure 2) inhibited reactive oxygen species (ROS) production.[17]

To further extend the structure-activity relationships (SARs) of derivatives 2 and to identify novel imidazo-pyrazoles endowed with antiproliferative/anti-inflammatory properties, we designed and synthesized a new library (twenty compounds) of imidazo-pyrazole compounds 3-5 ( Figure 2, Table 1). In the new derivatives we introduced the following chemical modifications: (i) decoration of the catechol ring with groups with different electronic, steric and lipophilic properties

Figure 2. Structure of new synthesized compounds 3-5.

Table 1. Decorations on the new synthesized compounds 3–5.					
Compd	R"	R'''	Yield [%]		
3 a	C <sub>6</sub> H <sub>5</sub>	CHF <sub>2</sub>	49		
3 b	$CH_2C_6H_5$	CHF <sub>2</sub>	50		
3 c	CH <sub>3</sub>	$CH_2C_6H_4-pF$	62		
3 d	CH₃	$CH_2C_6H_4$ -mF	47		
3 e	CH₃	$CH_2C_6H_4-pCI$	52		
4a	C₅H <sub>9</sub>	CH₃	68		
4 b	C₅H <sub>9</sub>	CHF <sub>2</sub>	68		
4 c	C <sub>6</sub> H <sub>5</sub>	CHF <sub>2</sub>	50		
4d	$CH_2C_6H_5$	CHF <sub>2</sub>	87		
4e	CH₃	$CH_2C_6H_4-pF$	92		
4f	CH₃	$CH_2C_6H_4$ -mF	50		
4g	CH <sub>3</sub>	CH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> -pCl	58		
5 a	$C_5H_9$	CH₃	88		
5 b	C₅H <sub>9</sub>	CHF <sub>2</sub>	42		
5 c	C <sub>6</sub> H <sub>5</sub>	CHF <sub>2</sub>	42		
5 d	$CH_2C_6H_5$	CHF <sub>2</sub>	79		
5 e	CH <sub>3</sub>	$CH_2C_6H_5$	40		
5 f	CH₃	$CH_2C_6H_4-pF$	62		
5 g	CH <sub>3</sub>	$CH_2C_6H_4$ -mF	84		
5 h	CH₃	CH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> -pCl	92		

(compounds 3); (ii) insertion of a methyl group on C-6 of imidazo-pyrazole scaffold (compounds 4); (iii) shift of the acylhydrazonic substituent from position 7 to position 6 of the imidazo—pyrazole substructure (compounds 5).

To confirm the biological activity of imidazo—pyrazole core, already observed in the previous compounds 1 and 2, we firstly submitted newly synthesized derivatives 3-5 to a preliminary screening to evaluate their antiproliferative action. Then, for the most promising molecules, we assessed cytotoxicity, ROS production inhibition and in silico pharmacokinetics and druglikeness properties. Finally, as previous imidazo-pyrazoles 2 a,b resulted active as tubulin polymerization inhibitors, [16] molecular docking and molecular dynamic simulations on tubulin system were carried out for the most potent antiproliferative compounds.

# **Results and Discussion**

#### Chemistry

The desired compounds 3-5 were synthesized through the four-step sequential procedure reported in Scheme 1.

In particular, the condensation of the proper ethyl cyanoacrylate 6a-c with 2-phenyl-1-hydrazinoethanol  $7^{\tiny{[18]}}$  led to the isolation of 3-amino pyrazoles 8a-c, as previously reported. [8,19] The dehydration of pyrazoles 8a-c with concentrated sulphuric acid (H<sub>2</sub>SO<sub>4</sub>) allowed the isolation of imidazo-pyrazole racemic esters 9a-c,[7,8] that were then transformed in the corresponding hydrazides 10a-c by the reaction with hydrazine hydrate (NH<sub>2</sub>NH<sub>2</sub>·H<sub>2</sub>O). Finally, the condensation between carbohydrazide intermediates 10 and the suitable benzaldehydes 11a-h (properly prepared according to literature methods)[17,20-23] led to the formation of the desired compounds 3-5 in good-tomoderate yields (Table 1).

Interestingly, the adopted reaction conditions for the formation of the acylhydrazone functionality proved to be stereoselective. In fact, although the condensation between compounds 10 and 11 could potentially lead to the formation of E/Z mixtures, only derivatives E were isolated, as assessed by NMR spectral analyses. As recently reported for similar hydrazones, [16,17] E-geometry was definitively assigned based on the chemical shift of the acylhydrazone proton atom. In particular, a chemical shift for CONH proton lower than 12 ppm (as in our case) was assigned to the E configuration, whereas  $\delta$ values higher than 15.5 ppm indicated a Z isomer. [24]

#### Antiproliferative activity

To confirm the anticancer activity of imidazo-pyrazole core, imidazo-pyrazoles 3-5 were tested at a fixed concentration of 10 μM against eight different cancer cell lines (namely, MCF7 breast adenocarcinoma, SKOV-3 ovarian adenocarcinoma, Hep-G2 hepatocellular carcinoma, SK-Mel28 cutaneous melanoma, MDA-MB-231 breast adenocarcinoma, SK-BR3 breast adenocar-

Scheme 1. Synthesis of compounds 3–5. Reagents and conditions: (a) Toluene, 70–80 °C, 8 h (8 a, 8 c) or absolute ethanol (abs. EtOH), reflux (8 b); (b) conc.  $H_2SO_4$ , 0 °C, 10 min., then concentrated ammonium solution (conc.  $NH_4OH$ ), r.t.; (c)  $NH_2NH_2 \cdot H_2O$ , 120–130 °C, 3–4 h; (d) abs. EtOH, 11 a–h, reflux 1–18 h.

cinoma, HeLa cervix epithelioid carcinoma and A549 lung carcinoma) using Cisplatin (Cis-Pt) as reference compound.

As reported in Table 2, selected compounds showed significative antiproliferative activity (cancer cell growth percentage <50% at 10  $\mu\text{M}$  concentration), resulting equally active or more effective than the reference compound. In detail, compounds 3, more strictly related to lead structures  $2\,a$ ,b (Figure 1), showed very interesting antiproliferative activity and derivatives  $3\,c$ ,  $3\,d$  and  $3\,e$ , sharing an halogen-substituted benzyloxy ring, potently inhibited the proliferation of A549, HeLa, SK-Mel28, MCF7 and SKOV-3 cell lines.

Derivatives 4, bearing an additional methyl group at C-6 position of the imidazo—pyrazole scaffold, resulted less active than compounds 3 against the considered panel of cell lines. However, compounds 4b and 4d selectively blocked cervix epithelioid carcinoma (HeLa cell line), whereas 4c shows inhibitory activity on HeLa, Hep-G2 and SK-Mel28 proliferation. Conversely, 4e broadly affect the proliferation of the considered cell lines.

Finally, among compounds **5**, bearing the acylhydrazone linker at position 6, **5f**, **5g** and **5h** showed interesting results (Table 2). In detail, **5f** was effective and selective towards hepatocellular carcinoma (Hep-G2 cell line), whereas **5g** exhibited strong action particularly against breast cancer cells MCF7 and MDA-MB2311. Compound **5h**, a close analogue of **2a** with an additional chlorine atom on benzyloxy substituent, strongly inhibited the proliferation of lung carcinoma (A549) and breast cancer (MBA-MB321) cells.

Collectively, the most potent imidazo—pyrazoles shared a difluoromethoxy substituents at para position of catechol moiety and a phenoxy group in meta position (derivatives 3a and 4c) or a 3-methoxy-4-substituted benzyloxy substructure (compounds 3e, 4e, 5g and 5h). In particular, the 4-chlorobenzyloxy substituted compounds 3e and 5g showed a broad spectrum of antiproliferative activity, whereas the presence of a smaller substituents (phenoxy and difuoromethoxy), as in compounds 3a and 4c, resulted in a more specific and selective anticancer action.

To further characterize the antiproliferative profile of the most active derivatives, the  $IC_{50}$  values of compounds **3 a**, **3 e**, **4 c**, **5 g** and **5 h** were determined on selected cancer cell lines (Table 3).

The tested compounds showed widespread and significant anticancer activity, thus confirming imidazo—pyrazole scaffold

Table 2. Cell growth percent values of imidazo-pyrazoles 3-5 and Cis-Pt used as reference compound on different cancer cell lines at  $10 \,\mu\text{M}$  concentration. For each compound, only cell lines with a growth percent values <50% are indicated. Data are mean values for three separate experiments. Variation among triplicate samples is less than 10%.

	Cell gro	owth [%]											
Cell line	3 a	3 c	3 d	3 e	4 b	4 c	4 d	4 e	5 e	5 f	5 g	5 h	Cis-Pt
A459 <sup>[a]</sup>	35.0	16.1	28.6	24.0	-	-	-	34.2	-	-	42.0	24.3	59.09
Hela <sup>[b]</sup>	34.9	16.7	22.1	31.9	37.5	44.7	41.0	22.4	45.3	-	39.9	33.8	29.33
Hep-G2 <sup>[c]</sup>	50.6	-	42.0	45.4	-	47.7	-	35.2	-	48.3	42.1	40.0	38.07
MCF7 <sup>[d]</sup>	34.0	17.4	23.3	49.9	-	-	-	27.9	-	_	33.6	_	72.74
MDA-MB321 <sup>[d]</sup>	33.3	35.4	49.7	42.4	-	-	-	39.5	-	_	31.8	26.0	86.07
SK-BR3 <sup>[d]</sup>	-	40.8	37.5	49.1	-	-	-	46.8	-	_	_	_	70.59
SK-Mel28 <sup>[e]</sup>	47.3	24.1	34.4	38.7	_	46.7	_	40.4	42.9	_	44.3	43.5	44.40
SKOV-3 <sup>[f]</sup>	45.4	19.8	32.7	28.6				37.4				39.9	36.83

[a] Lung carcinoma. [b] Cervix epithelioid carcinoma. [c] Hepatocellular carcinoma. [d] Breast cancer. [e] Cutaneous melanoma. [f] Ovarian adenocarcinoma.



$IC_{50}$ [ $\mu$ M] $\pm$ SD <sup>[g]</sup>						
Cell line	3 a	3 e	4 c	5 g	5 h	
A459 <sup>[a]</sup>	6.71 ± 0.13	2.74±0.38	-	_	2.57 ± 0.27	
Hela <sup>[b]</sup>	$6.30 \pm 0.59$	$4.72 \pm 0.06$	$30.16 \pm 3.64$	$21.67 \pm 0.56$	$6.30 \pm 0.59$	
Hep-G2 <sup>[c]</sup>	$\textbf{7.81} \pm \textbf{1.24}$	_	$21.49 \pm 0.61$	_	$7.10 \pm 0.12$	
MCF7 <sup>[d]</sup>	$7.57 \pm 0.37$	$7.37 \pm 1.10$	_	$22.01 \pm 1.83$	$7.67 \pm 0.54$	
MDA-MB321 <sup>[d]</sup>	$6.76 \pm 0.28$	$6.34 \pm 0.60$	_	$19.38 \pm 0.89$	$5.07 \pm 0.29$	
SK-BR3 <sup>[d]</sup>	$11.49 \pm 0.66$	_	_	_	$9.04 \pm 0.28$	
SK-Mel28 <sup>[e]</sup>	$5.49 \pm 0.48$	$3.08 \pm 0.10$	$30.60 \pm 1.72$	$18.28 \pm 0.92$	$4.94 \pm 0.41$	
SKOV-3 <sup>[f]</sup>	$7.36 \pm 0.28$	$2.95\pm1.03$	_	_	$6.41 \pm 0.28$	

[a] Lung carcinoma. [b] Cervix epithelioid carcinoma. [c] Hepatocellular carcinoma. [d] Breast cancer. [e] Cutaneous melanoma. [f] Ovarian adenocarcinoma.

as a valuable chemo-type for the development of novel antiproliferative agents. In particular,  ${\bf 3\,a}$  showed a IC50 value in the low micromolar range against SK-Mel 28, HeLa and A549 cancer cell lines (5.49 μM, 6.30 μM and 6.71 μM, respectively), while 3e was highly active against the ovarian adenocarcinoma (SKOV-3) and lung carcinoma (A549) cell lines (IC<sub>50</sub> values lower than  $3\,\mu\text{M}$ ). As anticipated by growth percentage values (Table 2), derivative 5 h showed a widespread activity (IC<sub>50</sub> values  $< 10 \,\mu\text{M})$  against all considered cell lines, being highly effective against lung carcinoma A549 ( $IC_{50} = 2.57 \mu M$ ), cutaneous melanoma SK-Mel 28 (IC $_{50}$ =4.94  $\mu$ M) and breast cancer MDA-MB231 (IC<sub>50</sub> = 5.02  $\mu$ M) lines. Noteworthy **3e** and **5h** are both characterized by a methoxy group in position 3 and the 4chloro-benzyloxy substituent in position 4 of the catechol ring, being chemically very similar to the lead compound 2b, previously resulted active as tubulin polymerization inhibitor.

Subsequently, the cytotoxic activity of the most potent compounds was assessed, on human embryonic fibroblasts (GM-6114) at 10  $\mu$ M concentration (Table S2, Supporting Information). Interestingly, derivatives  $\bf 3a$  and  $\bf 4c$  did not show any cytotoxicity, whereas compounds  $\bf 3e$ ,  $\bf 5g$  and  $\bf 5h$  evidenced some cell growth percent (50, 40.98 and 40.05% respectively), comparable to most common approved anticancer drug as Cis-Pt (39.52%).

# Inhibiting effect on human platelet aggregation and ROS production

Human platelets represent a common cellular element of thrombosis, inflammation, and cancer<sup>[25]</sup> and can be considered inflammatory cells, due to the release, during platelet activation/aggregation, of inflammatory cytokines, but even increasing the ROS concentration at the level of the vascular damage.<sup>[26]</sup> Therefore, platelets could be a simple, economic and suitable cellular model based on a causal relationship between inflammation and tumorigenesis.<sup>[2,27]</sup>

As previous imidazo—pyrazoles 2 blocked ROS production in platelet and to confirm anti-inflammatory activity of imidazo—pyrazole core already reported,<sup>[16,17]</sup> the inhibition of platelet aggregation and ROS production on human platelets were evaluated for the most active compounds 3a, 3e, 4c, 5g and 5h. As reported in Table 4, compounds 3a and 4c showed

**Table 4.** Inhibitory effect of new synthesized compounds **3a**, **3e**, **4c**, **5g**, **5h** on platelet aggregation and ROS production on human. Reported data are the mean +/- SD obtained in six different experiments each performed in duplicate.

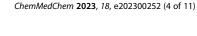
Compd	$IC_{50}\left[\mu M ight]\pm SD$ Aggregation Inhibition	ROS production Inhibition
3 a	85.20 ± 6.30	83.00 ± 22.40
3 e	$160.45 \pm 10.42$	179.24 ± 33.67
4 c	$83.89 \pm 3.36$	74.86 ± 6.18
5 g	$146.99 \pm 19.27$	173.22 ± 8.37
5 h	$140.57 \pm 15.85$	154.33 ± 0.38

the lower IC<sub>50</sub> values for platelet aggregation and ROS production inhibition, suggesting that the shift of the acylhydrazonic substituent from position 7 to position 6 (derivatives 5 g,h) or the insertion of bulky substituents on the catechol ring (derivative 3 e) have a detrimental effect on this activity. This specific trend agrees with previous results regarding compounds 2, where more embedded 2a and 2b (more similar to 3 e, 5 g and 5 h) resulted less active as ROS production inhibitors than 2 c, characterized by a smaller substituent on para position of catechol moiety.

#### Pharmacokinetic properties and drug likeness prediction

To evaluate the pharmaceutical relevance of the most active compounds, the pharmacokinetics and drug-likeness properties of derivatives 3a, 3e, 4c, 5g, 5h were calculated by SwissADME (Table S3, Supporting Information). Collectively the considered compounds are characterized by nine rotatable bonds, five to seven H-bond acceptors, two H-bond donors and TPSA values of 89.77 Ų, thus supporting a good capacity of all derivatives to permeate lipophilic barriers. In addition, all tested imidazo—pyrazoles were predicted to be well absorbed in gastrointestinal (GI) tract and are not expected to pass the brain-blood barrier (BBB); except 4c and 5h, they are moderately soluble in water. [29]

Derivatives **3a**, **3e**, **4c**, **5g**, **5h** would inhibit 2C19, 2C9, 2D6 and 3A4 cytochromes, without interacting with 1 A2 isoform. A Lipinski violation (MW > 500 Da) was identified for compounds **3e**, **4c** and **5h**, but none of the compounds showed any pan assay interference compound (PAINS) alert. Finally, the presence





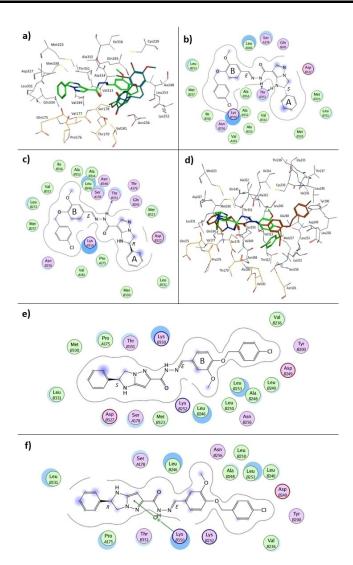
of a C=N functionality was identified as a limitation, according to the Brenk filter.  $^{\tiny{[30]}}$ 

#### **Docking studies**

As previous compounds **2** (particularly **2a** and **2b**) interfere with tubulin system, <sup>[16]</sup> the ability of new imidazo—pyrazoles **3e** and **5h** (close analogues of **2a** and **2b**) to bind the colchicine binding site in the polymeric complex tubulin  $\alpha$ /tubulin  $\beta$ / stathmin4 (PDB ID: 6XER)<sup>[31]</sup> was evaluated by docking simulations (AutoDock 4.2). <sup>[32]</sup> Both enantiomers of the two selected ligands were considered, whereas the stereochemistry of the hydrazone CH=N group was kept fixed in the *E* conformation, as assessed by NMR study.

Thus, in the S-3 e/tubulin docking complex (calculated Ki = 2.29 µM), the ligand would assume an extended conformation with the 4-chlorobenzyloxy portion located at equivalent position of the cyclohepta-2,4,6-trien-1-one moiety of colchicine (Figure 3a) and interacting with Lys350, Leu253, Asn256 and Met257 side chains. The complex would be mainly stabilized by two hydrogen bonds involving Thr351 backbone carbonyl and the imidazo-pyrazole and the carbohydrazide NH groups (Figure 3b). Furthermore, ring A (Figure 3b) would establish van der Waals contacts with Asp327 and the imidazo-pyrazole scaffold would interact with Ser178 (α-tubulin), Met323, Asp327 and Thr351 side chains. The interactions between the acylhydrazone linker and Ser178 (α-tubulin), Gln245, Lys350 and those occurring between ring B (Figure 3b) and Leu246 and Ala314 side chains would provide further stabilization to the complex. Isomer R-3e would assume a similar orientation within the binding site; however, the inversion of the stereocenter would affect the orientation of the imidazo-pyrazole portion and would prevent its ability to form a hydrogen bond with Thr351 (Figure 3c). The calculated Ki value for the R-3e/tubulin complex was 3.06 µM suggesting a decrease in affinity.

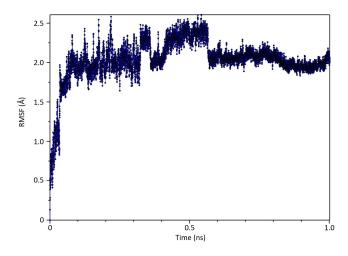
The shift of the acylhydrazone substituent from position 7 to position 6 of the imidazo-pyrazole ring would reduce the ability of compounds to bind tubulin. According to docking calculations, the Ki values for S-5 h/tubulin and R-5 h/tubulin complexes were 2- to 5-fold higher than those predicted for their  $3\,e$  analogues (Ki\_{(S-5h)}\!=\!12.72~\mu\text{M}; Ki\_{(R-5h)}\!=\!6.23~\mu\text{M}). The bioactive conformations predicted for  ${\bf 5}\,{\bf h}$  enantiomers followed those calculated for isomers 3e, except for the orientation of the 4-chlorobenzyloxy portion (Figure 3d). In particular, compound S-5h would adopt an elongated conformation with the 2-phenyl imidazo-pyrazole substructure in contact with Asp327, Leu331, Thr351, Lys350 and Pro175 ( $\alpha$ -tubulin) (Figure 3e). The phenyl ring B (Figure 3e) would interact with Leu246, Ala248, Lys252 and Asn256, whereas the 4-chlorobenzyloxy substructure would be in contact with Tyr200, Val236, Leu240, Asp249, Leu250 and Leu253. No hydrogen bond would occur between the ligand and tubulin dimer. In the R-5 h/ tubulin docking complex, the imidazo-pyrazole portion would form an arene-cation interaction with Lys350 side chain not observed in the S-5 h/tubulin complex (Figure 3f).



**Figure 3. a)** Superposition of the binding mode of *S*-**3 e** (docking pose, green) and colchicine (X-ray structure, dark green).  $\alpha$ - and  $\beta$ -Tubulin residues are coloured orange and grey, respectively. **b)** LigPlot of *S*-**3 e**/tubulin complex. **c)** LigPlot of *R*-**3 e**/tubulin complex. **d)** Superposition of the docking poses of *S*-**3 e** (green) and *S*-**5 h** (brown).  $\alpha$ - and  $\beta$ -Tubulin residues are coloured orange and grey, respectively. **e)** LigPlot of *R*-**5 h**/tubulin complex. **f)** LigPlot of *S*-**5 h**/tubulin complex.

# Molecular dynamics simulation

To further asses the reliability of the docking calculations, a molecular dynamic simulation (explicit solvent, T=310~K, time=1 ns) was carried out on the S-3 e/tubulin complex, selected as the most promising docking complex (calculated Ki=2.29  $\mu$ M). The analysis of the molecular dynamic trajectory highlighted a stable binding orientation for the ligand with small fluctuations form the docking pose (Figure 4). Furthermore, the analysis of the hydrogen bonding network between S-3 e and tubulin residues (occupancy value percentage) would confirm the interaction between the ligand and Thr351 side chain observed in the S-3 e/tubulin docking complex and would indicate the formation of two additional H-bond with Lys350 and Leu246.



**Figure 4.** Time dependent root-mean-square fluctuation (RMSF) analysis of ligand S-3 e.

## **Conclusions**

In this paper we investigated the pharmacological properties of imidazo-pyrazole scaffold, not well investigate, to provide important hits useful in medicinal chemistry research and to obtain new pharmacologically active compounds. The SAR extension study on compounds 1 and 2, allowed the identification of imidazo-pyrazoles 3-5 endowed with significative antiproliferative and antioxidant properties. Particularly, compounds 3, more strictly related to lead structures 2 a,b, showed very interesting antiproliferative activity and collectively the most promising compounds (3 a, 3 e, 4 c, 5 g and 5 h) resulted no cytotoxic on healthy cells and significantly inhibit ROS production in platelet. In addition, they are endowed with good in silico pharmacokinetics and drug-likeness properties. Finally, molecular docking and molecular dynamic simulations confirmed the possible interaction of most active compounds 3e and 5h with tubulin, identified as the biological target for lead derivatives 2a,b. Collectively, these biological results support the pharmacological potential of this class of derivatives as novel therapeutic agents.

The collected SARs indicate that the antiproliferative activity is affected by both substitutions on catechol portion and the decoration of imidazo-pyrazole scaffold. Thus, the insertion of an acylhydrazonic moiety at position 6 or 7 led to potent antiproliferative agents (namely, compounds 3a,e, 4c, 5g,h) whose tumour-specificity is influenced by the substituents of the catechol ring. Compounds bearing bulky substituents (i.e., substituted-benzyloxy and phenoxy) at para position of catechol moiety (as 3e, 5g and 5h), as well as derivatives with the same substituents in meta position (as 3a and 4c), provided interesting results. Thus, compounds 3e and 5g, sharing the para-chlorobenzyloxy moiety, resulted the most effective in conferring a broad spectrum of antiproliferative activity and showed a potent action particularly against lung carcinoma A459 cell lines, with low micromolar IC<sub>50</sub> values. On the contrary, the presence of a smaller substituents (phenoxy and difuoromethoxy), as in compounds  $\bf 3a$  and  $\bf 4c$ , resulted in a more specific and selective anticancer action and in a great antioxidant potency. In fact, some compounds (i.e.,  $\bf 3e$  and  $\bf 5h$ ) inhibited the proliferation of several tumor cell lines, while other derivatives (i.e.,  $\bf 4b$ ,  $\bf 4d$  and  $\bf 5f$ ) blocked the growth of specific tumors (namely, HeLa and Hep-G2) worthy of further investigation. On the contrary, as previously observed for derivatives  $\bf 2$ , the antioxidant activity is increased when a poorly hindered substituents (as OCHF2) are present at para position of catechol portion (see derivatives  $\bf 3a$  and  $\bf 4c$ ).

All these collected results confirmed that imidazo—pyrazole scaffold linked to a catechol portion differently decorated could represent a valuable moiety for the development of new anticancer/anti-inflammatory agents able to act at the intracellular level with different pathways, as ROS production inhibition and tubulin system.<sup>[33-35]</sup> In particular, derivatives 3 were identified as promising agents endowed with antiproliferative and antioxidant dual activity. As previously demonstrated, these multiple activities of imidazo—pyrazole chemotype could be partially explained by the close correlation between different intracellular pathways, as tubulin system and ROS production, involved in cancer progression and inflammation.<sup>[37,38]</sup>

In conclusion, these data may be of interest for the development of future new and more active imidazo—pyrazoles able to interfere with cancer and inflammation with different intracellular mechanisms.

# **Experimental Section**

# Chemistry

All chemicals were purchased from Chiminord and Aldrich Chemical (Milan, Italy). Solvents were reagent grade. All commercial reagents were used without further purification. Aluminium backed silica gel plates (Merck DC-Alufolien Kieselgel 60 F254, Darmstadt, Germany), were used in thin-layer chromatography (TLC) for routine monitoring the reaction course. Detection of spots was made by UV light. Merck silica gel, 230-400 mesh, was used for chromatography. Flash chromatography was performed using Isolera one instrument (Biotage, Uppsala, Sweden), using a silica gel column. Melting points are not "corrected" and were measured with a Buchi M-560 instrument (Buchi instruments, Flawil, Switzerland). IR spectra were recorded with a Perkin-Elmer 398 spectrophotometer (Perkin-Elmer, Milan, Italy). <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on JEOL JNM-ECZR (Tokyo, Japan) instrument or a Varian Gemini 200 (200 MHz, Varian Gemini, Palo Alto, CA, USA) using deuterochloroform (CDCl<sub>3</sub>) or dimethyl sulfoxide (DMSO-d<sub>6</sub>) as solvent; chemical shifts are reported as  $\delta$  (ppm) and signals were characterized as s (singlet), d (doublet), t (triplet), nt (near triplet), q (quartet), m (multiplet), br s (broad signal); J are reported in Hz. IR spectra were recorded with a Perkin-Elmer 398 spectrophotometer (Perkin-Elmer, Milan, Italy).

Elemental analysis was determined with an elemental analyzer EA 1110 (Fison-Instruments, Milan, Italy) and the purity of all synthesized compounds was >95%; products are considered pure when the difference between calculated and found values is  $\pm$  than 0.4.

Compounds  $8\,a$ ,c,  $9\,a$ ,c and  $10\,a$  have been synthesized as previously reported.  $^{[7,8,16,17]}$ 



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# Synthesis of 2-hydrazino-1-phenylethanol 7<sup>[18]</sup>

To  $NH_2NH_2 \cdot H_2O$  (30 mL, 0.6 mol) at  $100\,^{\circ}C$ , styrene oxide (20 mL, 0.17 mol) is added dropwise, and the mixture is heated at  $100\,^{\circ}C$  for 1 hour.

Then the excess of hydrazine is removed under reduced pressure and the residue is purified by distillation under high vacuum to obtain a colorless liquid used as crude without purification. Molecular formula:  $C_8H_{12}N_2O$  (M.W.: 152.19). Yield: 87%. B.p.: 155–158 °C/0.6 mmHg.

# Synthesis of ethyl 5-amino-1-(2-hydroxy-2-phenylethyl)-3-methyl-1*H*-pyrazole-4-carboxylate 8 b

To 2-cyano-3-ethoxybut-2-enoate<sup>[39]</sup> (1.2 g, 6 mmol) solved in abs. EtOH (5 mL), 2-hydrazino-1-phenylethanol **10** (0.92 g, 6 mmol) is added and the mixture is heated for 8 hours. After cooling to r.t., solvent is removed under reduced pressure. The brown oil obtained is crystallized from ethyl ether to obtain a white solid. Yield: 57%. M.p.: 123–125 °C. IR (CHCl<sub>3</sub>): 3440, 3355 (NH<sub>2</sub>), 1670 (CO) cm<sup>-1</sup>. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  1.13 (t, J=7.0 Hz, 3H,  $CH_3$ CH<sub>2</sub>), 2.40 (s, 3H, CH<sub>3</sub>), 3.80–4.38 (m, 4H, CH<sub>2</sub>O+CH<sub>2</sub>N), 4.93–5.08 (m, 2H, CHOH+OH), 5.43 (br s, 2H, NH<sub>2</sub>), 7.02–7.11 (m, 5H, Ar). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  162.63, 153.86, 150.94, 142.39, 128.56, 128.19, 127.25, 97.66, 73.02, 60.04, 56.32, 14.93, 14.25. Elemental Analysis calculated for C<sub>15</sub>H<sub>19</sub>N<sub>3</sub>O<sub>3</sub>.

# Synthesis of ethyl 6-methyl-2-phenyl-2,3-dihydro-1*H*-imidazo[1,2-*b*]pyrazole-7-carboxylate 9 b

Compound **8 b** (2.9 g, 10 mmol) is solved in conc.  $H_2SO_4(10 \text{ mL})$  at 0 °C and the mixture is stirred at r.t. for 15 minutes. Then, iced  $H_2O$  (500 mL) is added and the solution is made neutral with conc.  $NH_4OH$  solution; the white solid obtained is filtered, washed with  $H_2O$  and recrystallized from abs. EtOH. Yield: 80 %. M.p.: 140–141.5 °C. IR (KBr): 3333 (NH), 1661 (CO) cm<sup>-1</sup>.  $^1H$ -NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  1.21–1.43 (m, 3H,  $CH_3CH_2$ ), 2.44 (s, 3H,  $CH_3$ ), 3.87–4.03 (nt, 1H, H3), 4.16–4.37 (m, 2H,  $CH_3CH_2$ ), 4.43–4.60 (nt, 1H, H3), 5.00 (br, 1H, NH disappear with  $D_2O$ ), 5.33–5.47 (m, 1H, H2), 7.24–7.51 (m, 5H, Ar).  $^1$ 3C-NMR (101 MHz, CDCl<sub>3</sub>): 162.08, 157.17, 151.37, 140.65, 128.58, 127.72, 126.02, 92.36, 67.37, 60.04, 57.85, 15.14, 14.25. Elemental Analysis calculated for  $C_{15}H_{17}N_3O_2$ .

# General procedure for synthesis of compounds 10 a-c

Compounds **9a–c** (15 mmol) and  $NH_2NH_2 \cdot H_2O$  (14 mL, 750 mg, 16 mmol) are heated to 120–130 °C for 3 hours. After cooling to r.t.,  $H_2O$  (50 mL) is added and the white solid obtained is filtered, washed several times with  $H_2O$  and recrystallized from 95 % EtOH.

6-Methyl-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (10 b). Yield: 51 %. M.p.: 231–232 °C.  $^1$ H-NMR (400 MHz, DMSO-d<sub>6</sub>): 2.27 (s, 3H, CH<sub>3</sub>), 3.70 (nt, 1H, H3), 4.23 (br s, 2H, NH<sub>2</sub> disappears with D<sub>2</sub>O), 4.49 (nt, 1H, H3), 5.36 (nt, 1H, H2), 6.99 (br s, 1H, NH disappears with D<sub>2</sub>O), 7.31–7.44 (m, 5H, Ar), 8.11 (s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>): δ 166.61, 157.30, 146.65, 140.79, 128.58, 127.72, 126.02, 90.02, 67.34, 58.00, 15.13. Elemental Analysis calculated for C<sub>13</sub>H<sub>15</sub>N<sub>5</sub>O.

2-Phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide ( $10\,c$ ). Yield:76 %. M.p. 199 – 200 °C.  $^1$ H-NMR (400 Mz, DMSO-d<sub>6</sub>): 3.81 (nt, 1H, H3), 4.31 (br s, 2H, NH<sub>2</sub>, disappears with D<sub>2</sub>O), 4.60 (nt, 1H, H3), 5.35 (nt, 1H, H2), 5.67 (s, 1H, H7), 6.57 (s, 1H, NH, disappears with D<sub>2</sub>O), 7.30–7.45 (m, 5H, Ar), 9.13 (s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  162.85, 154.75, 146.27,

140.17, 128.58, 127.72, 126.02, 82.04, 67.85, 56.63. Elemental Analysis calculated for  $C_{12}H_{13}N_5O$ .

#### General procedure for synthesis of compounds 3-5

To a solution of carbohydrazide 10a-c (1 mmol) in abs. EtOH (10 mL) the suitable aldehyde 6a-g (1 mmol) solved in abs. EtOH (2 mL) is added dropwise. The reaction mixture is heated at reflux for 1–18 hours. After cooling to room temperature, the solvent is removed under reduced pressure to obtain yellow/white solids which are filtered and re-crystallized from abs. EtOH or ethyl acetate.

(E)-N'-(4-(difluoromethoxy)-3-phenoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (3 a). M.p: 99–100 °C (abs. ethanol).  $^1\text{H-NMR}$  (200 MHz, DMSO-d<sub>6</sub>):  $\delta$  3.75 (nt, 1H, H3), 4.55 (nt, 1H, H3), 5.38 (nt, 1H, H2), 6.93–7.80 (m, 15H, 10Ar+H-5 Ar+H-6 Ar+H-2 Ar+OCHF $_2$ +H6), 8.00 (br s, 1H, CH=N), 11.14 (s, 1H, NH disappears with D2O).  $^{13}\text{C-NMR}$  (100 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.08, 156.83, 153.01, 147.67, 145.13, 144.77, 137.24, 130.02, 128.58, 128.30, 127.72, 126.02, 124.60, 122.31, 118.78, 115.76, 114.87, 113.66, 95.14, 67.31, 57.64. Elemental Analysis calculated for  $C_{26}H_{21}F_{2}N_{5}O_{3}$ .

(*E*)-*N'*-(3-(benzyloxy)-4-(difluoromethoxy)benzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (3 b). M.p. 134–135 °C (ethyl acetate). <sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>): δ 3.82 (nt, 1H, H3), 4.62 (nt, 1H, H3), 5.43 (m, 2H, CH<sub>2</sub>O), 5.46 (nt, 1H, H2), 6.97–7.68 (m, 15H, 10Ar+H-5 Ar+H-6 Ar+H-2 Ar+OCHF<sub>2</sub>+H6), 7.94 (s, 1H, CH=N), 11.19 (s, 1H, NH disappears with D<sub>2</sub>O). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>): δ 160.08, 153.01, 150.58, 144.67, 140.79, 137.49, 128.56, 128.49, 128.20, 128.15, 127.72, 126.02, 122.14, 118.96, 117.87, 115.77, 113.66, 111.68, 95.14, 71.24, 67.31, 57.64. Elemental Analysis calculated for  $C_{27}H_{23}F_2N_5O_3$ .

(E)-N'-(4-((4-fluorobenzyl)oxy)-3-methoxybenzylidene)-2-phenyl-2,3-di-hydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (3 c). M.p: 119–120 °C (abs. ethanol).  $^1\text{H-NMR}$  (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  3.79–3.84 (m, 4H, OCH $_3$ +H3), 4.61 (nt, 1H, H3), 5.11 (s, 2H, CH $_2$ O), 5.45 (nt, 1H, H2), 7.08–7.52 (m, 13H, 10Ar+H-5 Ar+H-6 Ar+H-2 Ar+H6), 7.98 (br s, 1H, CH=N), 11.06 (br s, 1H, CONH, disappears with D $_2$ O).  $^{13}\text{C-NMR}$  (101 MHz, DMSO-d $_6$ ):  $\delta$  163.56, 161.59, 160.08, 153.01, 150.54, 144.53, 140.79, 137.24, 133.38, 130.14, 130.07, 128.58, 127.72, 126.02, 121.64, 115.46, 115.28, 114.35, 109.98, 95.14, 71.29, 67.31, 57.64, 56.02. Elemental Analysis calculated for  $\text{C}_{27}\text{H}_{24}\text{FN}_5\text{O}_3$ .

(E)-N'-(4-((3-fluorobenzyl)oxy)-3-methoxybenzylidene)-2-phenyl-2,3-di-hydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (3 d). M.p: 191–192 °C (abs. ethanol). H-NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  3.79–3.85 (m, 4H, OCH $_3$  + H3), 4.61 (nt, 1H, H3), 5.16 (s, 2H, CH $_2$ O), 5.45 (nt, 1H, H2), 7.07–7.48 (m, 13H, 9Ar + H-5 Ar + H-6 Ar + H-2 Ar + H6), 7.98 (br s, 1H, CH=N), 11.07 (br s, 1H, CONH, disappears with D $_2$ O).  $^{13}$ C-NMR (101 MHz, DMSO-d $_6$ ):  $\delta$  162.51, 160.54, 1, 153.01, 150.57, 144.53, 140.79, 138.05, 137.24, 130.01, 129.93, 128.58, 127.72, 126.02, 122.98, 121.64, 114.37, 114.18, 114.05, 113.87, 109.96, 95.14, 70.46, 67.31, 57.64, 56.02. Elemental Analysis calculated for  $C_{27}H_{24}FN_5O_3$ .

(*E*)-*N'-*(4-((4-chlorobenzyl)oxy)-3-methoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (3 e). M.p. 118–119 °C (abs. ethanol). ¹H-NMR (400 MHz, DMSO-d<sub>6</sub>): δ 3.79–4.07 (m, 4H, OCH<sub>3</sub> + H3), 4.61 (nt, 1H, H3), 5.13 (s, 2H, OCH<sub>2</sub>), 5.45 (nt, 1H, H2), 7.06–7.50 (m, 13H, 9Ar + H-5 Ar + H-6 Ar + H-2 Ar + H6), 7.98 (br s, 1H, CH=N), 11.06 (br s, 1H, CONH, disappears with D<sub>2</sub>O). ¹³C-NMR (101 MHz, DMSO-d<sub>6</sub>): δ 160.08, 153.01, 150.54, 144.53, 140.79, 137.24, 135.25, 134.08, 129.75, 128.64, 128.58, 127.72, 126.02, 121.64, 114.35, 109.98, 95.14, 71.29, 67.31, 57.64, 56.02. Elemental Analysis calculated for  $C_{27}H_{24}CIN_5O_3$ .



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(E)-N'-(3-(cyclopentyloxy)-4-methoxybenzylidene)-6-methyl-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4a). M.p: 151–152 °C (abs. ethanol).  $^{1}$ H-NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  1.46–1.68 (m, 8H, 4CH<sub>2</sub> cyclopent.), 2.47 (s, 3H, CH<sub>3</sub>), 4.04–4.08 (m, 4H, OCH<sub>3</sub>+ H3), 4.52 (nt, 1H, H3), 4.84-4.85 (m, 1H, CHO cyclopent.), 5.27 (nt, 1H, H2), 5.51 (br s, 1H, NH, disappears with  $D_2O$ ), 6.99 (d, J=8 Hz, 1H, H-5 Ar), 7.10 (d, J = 8 Hz, 1H, H-6 Ar), 7.18 -7.54 (m, 6H, 5 Ar +H-2 Ar), 7.97 (s, 1H, CH=N), 11.07 (br s, 1H, CONH, disappears with  $D_2O$ ). <sup>13</sup>C-NMR (101 MHz, DMSO- $d_6$ ):  $\delta$  160.35, 152.32, 151.83, 147.74, 145.85, 144.57, 140.79, 128.58, 128.06, 127.72, 126.02, 124.23, 113.06, 112.85, 91.10, 80.51, 67.34, 58.00, 56.09, 32.81, 23.45, 15.11. Elemental Analysis calculated for C<sub>26</sub>H<sub>29</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-(cyclopentyloxy)-4-(difluoromethoxy)benzylidene)-6-methyl-2phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4b). M.p: 110–111 °C (abs. ethanol).  $^{1}$ H-NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$ 1.59-1.97 (m, 8H, 4CH<sub>2</sub> cyclopent.), 2.46 (s, 3H, CH<sub>3</sub>), 3.81 (nt, 1H, H3), 4.61 (nt, 1H, H3), 4.90-4.93 (m, 1H, CHO cyclopent.), 5.42 (nt, 1H, H2), 7.01 (t, J=70 Hz, 1H, OCHF<sub>2</sub>), 7.13–7.42 (m, 8H, 5Ar+H-5Ar + H-6 Ar + H-2 Ar), 7.99 (br s, 1H, CH=N), 11.17 (br s, 1H, CONH, disappears with  $D_2O$ ).  $^{13}C$ -NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.35, 152.32, 149.05, 145.85, 145.20, 145.15, 144.67, 140.79, 128.58, 128.00, 127.72, 126.02, 121.92, 119.28, 117.87, 115.77, 113.66, 91.10, 81.11, 67.34, 58.00, 32.81, 23.45, 15.11. Elemental Analysis calculated for  $C_{26}H_{27}F_2N_5O_3$ .

(E)-N'-(4-(difluoromethoxy)-3-phenoxybenzylidene)-6-methyl-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (**4c**). <sup>1</sup> M.p: 130–131 °C (ethyl acetate). H-NMR (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  2.58 (s, 3H, CH<sub>3</sub>), 4.20 (nt, 1H, H3), 4.64 (nt, 1H, H3), 5.48 (nt, 1H, H2), 7.01-7.40 (m, 17H, 13Ar + H-5 Ar + H-6 Ar + H-2 Ar + OCHF<sub>2</sub>), 8.03 (s, 1H, CH=N), 11.16 (br s, 1H, CONH, disappears with D<sub>2</sub>O). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.35, 156.83, 152.32, 147.66, 145.85, 145.08, 144.77, 140.79, 130.02, 128.58, 128.30, 127.72, 126.02, 124.60, 122.31, 118.76, 117.87, 115.76, 114.87, 113.66, 91.10, 67.34, 58.00, 15.11. Elemental Analysis calculated for C<sub>27</sub>H<sub>23</sub>F<sub>2</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-(benzyloxy)-4-(difluoromethoxy)benzylidene)-6-methyl-2phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4d). M.p: 129–131 °C (ethyl acetate).  $^{1}$ H-NMR (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  2.46 (s, 3H, CH<sub>3</sub>), 3.79 (nt, 1H, H3), 4.59 (nt, 1H, H3), 5.20 (s, 2H, OCH<sub>2</sub>), 5.42 (nt, 1H, H2), 6.93-7.46 (m, 14H, 10Ar+H-5 Ar+H-6 Ar+H-2 Ar +OCHF<sub>2</sub>), 7.91 (br s, 1H, CH=N), 11.15 (br s, 1H, CONH, disappears with D $_{2}\text{O}\text{)}.$   $^{13}\text{C-NMR}$  (101 MHz, DMSO-d $_{6}\text{)}:$   $\delta$  160.35, 152.32, 150.56, 145.85, 144.54, 140.79, 137.49, 128.58, 128.22, 128.16, 127.72,, 126.02, 122.14, 118.98, 117.87, 115.77, 113.66, 111.68, 91.10, 71.24, 67.34, 58.00, 15.11. Elemental Analysis calculated for  $C_{28}H_{25}F_2N_5O_3$ .

(E)-N'-(4-((4-fluorobenzyl)oxy)-3-methoxybenzylidene)-6-methyl-2phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4e). M.p: 78–80  $^{\circ}$ C (abs. ethanol).  $^{1}$ H-NMR (400 MHz DMSO-d<sub>6</sub>):  $\delta$  2.47 (s, 3H, CH<sub>3</sub>), 3.32 (s, 3H, OCH<sub>3</sub>), 3.86 (nt, 1H, H3), 4.45 (nt, 1H, H3), 5.05 (br s, 1H, NH disappears with D2O), 5.10 (s, 2H, OCH2), 5.49 (nt, 1H, H2), 7.22-7.55 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.63 (s, 1H, CH=N), 11.19 (br s, 1H, CONH, disappears with D<sub>2</sub>O). <sup>13</sup>C-NMR (101 MHz, DMSO- $d_6$ ):  $\delta$  163.56, 161.59, 160.35, 152.32, 150.54, 145.85, 144.53, 140.79, 133.38, 130.14, 130.07, 128.58, 127.72, 126.02, 121.64, 115.46, 115.28, 114.35, 109.98, 91.10, 71.29, 67.34, 58.00, 56.02, 15.11. Elemental Analysis calculated for C<sub>28</sub>H<sub>26</sub>FN<sub>5</sub>O<sub>3</sub>.

(E)-N'-(4-((3-fluorobenzyl)oxy)-3-methoxybenzylidene)-6-methyl-2phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4f). M.p: 98–99 °C (abs. ethanol).  $^{1}$ H-NMR (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  2.47 (s, 3H, CH<sub>3</sub>), 3.41 (s, 3H, OCH<sub>3</sub>), 3.76 (nt, 1H, H3), 4.45 (nt, 1H, H3), 5.07 (s, 2H, OCH<sub>2</sub>), 5.35 (nt, 1H, H2), 5.44 (br s, 1H, NH disappears with  $D_2O$ ), 6.92–7.40 (m, 12H, 9Ar + H-5 Ar + H-6 Ar + H-2 Ar), 8.14 (s, 1H, CH=N), 11.19 (br s, 1H, CONH, disappears with D<sub>2</sub>O). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  162.51, 160.54, 160.35, 152.32, 150.57, 145.85, 144.53, 140.79, 138.05, 137.99, 130.01, 129.93, 128.58, 127.72, 126.02, 122.95, 121.64, 114.37, 114.18, 113.87, 109.96, 91.10, 70.46, 67.34, 58.00, 56.02, 15.11. Elemental Analysis calculated for  $C_{28}H_{26}FN_5O_3$ .

(E)-N'-(4-((4-chlorobenzyl)oxy)-3-methoxybenzylidene)-6-methyl-2phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-7-carbohydrazide (4g). M.p: 91–92 °C (abs. ethanol).  $^{1}$ H-NMR (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  2.47 (s, 3H, CH<sub>3</sub>), 3.46 (s, 3H, OCH<sub>3</sub>), 3.73 (nt, 1H, H3), 4.46 (nt, 1H, H3), 5.07 (s, 2H, OCH<sub>2</sub>), 5.27–5.30 (m, 2H, H2+NH, 1H disappears with  $D_2O$ ), 7.17-7.50 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.01 (s, 1H, CH=N), 11.19 (br s, 1H, CONH, disappears with D<sub>2</sub>O). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.35, 152.32, 150.54, 145.85, 144.53, 140.79, 135.25, 134.08, 129.75, 128.64, 128.58, 127.72, 126.02, 121.64, 114.35, 109.98, 91.10, 71.29, 67.34, 58.00, 56.02, 15.11. Elemental Analysis calculated for C<sub>28</sub>H<sub>26</sub>CIN<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-(cyclopentyloxy)-4-methoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide (5 a). M.p: 174-175 °C (ethyl acetate).  $^{1}$ H-NMR (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  1.64–1.69 (m, 8H, 4CH<sub>2</sub> cyclopent.), 3.85 (s, 3H, OCH<sub>3</sub>), 3.87 (nt, 1H, H3), 4.64 (nt, 1H, H3), 4.65-4.67 (m, 1H, CHO cyclopent.), 5.35 (nt, 1H, H2), 5.77 (s, 1H, H7), 6.62 (s, 1H, NH, disappears with  $D_2O$ ), 6.99 (d, J=8 Hz, 1H, H-5 Ar), 7.09 (d, J = 2 Hz, 1H, H-6 Ar), 7.15 (s, 1H, H-2 Ar), 7.26–7.44 (m, 5H, Ar), 8.29 (s, 1H, CH=N), 11.18 (br s, 1H, CONH, disappears with  $D_2O$ ). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.84, 154.76, 151.83, 147.74, 145.70, 144.69, 140.17, 128.58, 128.06, 127.72, 126.02, 124.28, 113.06, 112.85, 83.28, 80.51, 67.86, 56.63, 56.09, 32.81, 23.45. Elemental Analysis calculated for C<sub>25</sub>H<sub>27</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-(cyclopentyloxy)-4-(difluoromethoxy) benzylidene)-2-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-phenyl-pheny*2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide* (*5 b*). M.p: 145–149 °C (ethyl acetate).  $^{1}$ H-NMR (400 MHz DMSO-d<sub>6</sub>):  $\delta$  1.50–1.68 (m, 8H, 4CH<sub>2</sub> cyclopent.), 3.96 (nt, 1H, H3), 4.48 (nt, 1H, H3), 4.88-4.90 (m, 1H, CHO cyclopent.), 5.31 (nt, 1H, H2), 5.73 (s, 1H, H7), 6.75 (br s, 1H, NH, disappears with  $D_2O$ ), 7.22 (t, J=70 Hz, 1H,  $OCHF_2$ ), 7.26–7.38 (m, 8H, H-5 Ar + H-6 Ar + H-2 Ar + 5Ar), 7.94 (s, 1H, CH= $\mathbb{N}$ ), 11.18 (br s, 1H, CONH, disappears with  $D_2O$ ).  $^{13}C\text{-NMR}$  (101 MHz, DMSO-d<sub>6</sub>): δ 160.84, 154.76, 149.03, 145.70, 145.15, 144.66, 140.17, 128.58, 127.96, 127.72, 126.02, 121.95, 119.30, 119.28, 117.87, 115.77, 113.66, 112.99, 83.28, 81.11, 67.86, 56.63, 32.81, 23.45. Elemental Analysis calculated for C<sub>25</sub>H<sub>25</sub>F<sub>2</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(4-(difluoromethoxy)-3-phenoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide (5 c). M.p: 168-169 °C (ethyl acetate).  $^{1}\text{H-NMR}$  (200 MHz,DMSO-d<sub>6</sub>):  $\delta$  3.51 (br s, 1H, NH, disappears with D<sub>2</sub>O), 3.84 (nt, 1H, H3), 4.60 (nt, 1H, H3), 5.39 (nt, 1H, H2), 5.82 (s, 1H, H7), 6.98-7.40 (m, 14H, 10Ar+H-5 Ar+H-6  $Ar + H-2 Ar + OCHF_2$ ), 8.69 (s, 1H, CH=N), 11.18 (br s, 1H, CONH, disappears with  $D_2O$ ). <sup>13</sup>C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.84, 156.83, 154.76, 147.66, 145.70, 145.08, 144.66, 140.17, 130.02, 128.58, 128.32, 127.72, 126.02, 124.60, 122.31, 118.78, 117.87, 115.76, 114.87, 113.66, 83.28, 67.86, 56.63. Elemental Analysis calculated for  $C_{26}H_{21}F_2N_5O_3$ .

(E)-N'-(3-(benzyloxy)-4-(difluoromethoxy)benzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide (5 d). M.p: 96-98 °C (abs. ethanol). <sup>1</sup>H-NMR (200 MHz,DMSO-d<sub>6</sub>): δ 3.90 (nt, 1H, H3), 4.70 (nt, 1H, H3), 5.22 (s, 2H, OCH<sub>2</sub>), 5.42 (nt, 1H, H2), 5.73 (s, 1H, H7), 6.70 (br s, 1H, NH, disappears with  $D_2O$ ), 7.11–7.57 (m, 13H, H-5  $Ar + H-6 Ar + H-2 Ar + OCHF_2 + 9Ar)$ , 8.39 (s, 1H, CH=N), 11.16 (br s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$ 160.84, 154.76, 150.56, 145.70, 144.66, 144.49, 140.17, 137.49,  $128.58,\ 128.18,\ 127.72,\ 126.02,\ 122.14,\ 118.98,\ 117.87,\ 115.77,$ 113.66, 111.68, 83.28, 71.24, 67.86, 56.63. Elemental Analysis calculated for C<sub>27</sub>H<sub>23</sub>F<sub>2</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(4-(benzyloxy)-3-methoxybenzylidene)-2-phenyl-2,3-dihydro-1Himidazo[1,2-b]pyrazole-6-carbohydrazide (5 e). M.p: 99-100°C (ethyl





acetate).  $^{1}$ H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  3.81 (s, 3H, OCH<sub>3</sub>), 3.87 (nt, 1H, H3), 4.68 (nt, 1H, H3), 5.10 (s, 2H, OCH<sub>2</sub>), 5.36 (nt, 1H, H2), 5.75 (s, 1H, H7), 6.68 (br s, 1H, NH, disappears with D2O), 7.02–7.49 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.36 (s, 1H, CH=N), 11.45 (br s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.84, 154.76, 150.54, 150.52, 145.70, 144.69, 140.17, 136.03, 128.58, 128.16, 127.85, 127.72, 127.71, 126.02, 121.64, 114.35, 109.98, 83.28, 71.06, 67.86, 56.63, 56.02. Elemental Analysis calculated for C<sub>27</sub>H<sub>24</sub>N<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-((4-fluorobenzyl)oxy)-4-methoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide (5 f). M.p.: 175–176 °C (abs. ethanol).  $^1$ H-NMR (400 MHz, DMSO-d<sub>6</sub>): δ 3.79 (s, 3H, OCH<sub>3</sub>), 3.87 (nt, 1H, H3), 4.62 (nt, 1H, H3), 5.06 (s, 2H, OCH<sub>2</sub>), 5.32 (nt, 1H, H2), 5.73 (s, 1H, H7), 6.59 (br s, 1H, NH, disappears with D<sub>2</sub>O), 7.03–7.55 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.38 (s, 1H, CH=N), 11.18 (br s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>): δ 163.56, 161.59, 160.84, 154.76, 150.54, 145.70, 144.69, 140.17, 133.38, 130.14, 130.07, 128.58, 127.85, 127.72, 126.02, 121.64, 115.46, 115.28, 114.35, 109.98, 83.28, 71.29, 67.86, 56.63, 56.02. Elemental Analysis calculated for C<sub>27</sub>H<sub>24</sub>FN<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-((3-fluorobenzyl)oxy)-4-methoxybenzylidene)-2-phenyl-2,3-di-hydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide ( $\mathbf{5g}$ ). M.p.: 193–194 °C (abs. ethanol).  $^1$ H-NMR (400 MHz,DMSO-d<sub>6</sub>): δ 3.80 (s, 3H, OCH<sub>3</sub>), 3.90 (nt, 1H, H3), 4.66 (nt, 1H, H3), 5.06 (s, 2H, OCH<sub>2</sub>), 5.35 (nt, 1H, H2), 5.70 (s, 1H, H7), 6.61 (br s, 1H, NH, disappears with D<sub>2</sub>O), 7.03–7.41 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.30 (s, 1H, CH=N), 11.18 (br s, 1H, CONH, disappears with D<sub>2</sub>O).  $^{13}$ C-NMR (101 MHz, DMSO-d<sub>6</sub>): δ 162.51, 160.84, 154.76, 150.57, 145.70, 144.69, 140.17, 138.05, 130.01, 129.93, 128.58, 127.85, 126.02, 122.98, 121.65, 114.37, 114.18, 114.05, 113.87, 109.96, 83.28, 70.46, 70.43, 67.86, 56.63, 56.02. Elemental Analysis calculated for  $C_{27}$ H<sub>24</sub>FN<sub>5</sub>O<sub>3</sub>.

(E)-N'-(3-((4-chlorobenzyl)oxy)-4-methoxybenzylidene)-2-phenyl-2,3-dihydro-1H-imidazo[1,2-b]pyrazole-6-carbohydrazide (5 h). M.p.: 204–205 °C (abs. ethanol).  $^1\text{H-NMR}$  (400 MHz,DMSO-d<sub>6</sub>):  $\delta$  3.76 (s, 3H, OCH<sub>3</sub>), 3.85 (nt, 1H, H3), 4.59 (nt, 1H, H3), 5.07 (s, 2H, OCH<sub>2</sub>), 5.39 (nt, 1H, H2), 5.74 (s, 1H, H7), 6.62 (br s, 1H, NH, disappears with D<sub>2</sub>O), 6.97–7.53 (m, 12H, 9Ar+H-5 Ar+H-6 Ar+H-2 Ar), 8.27 (s, 1H, CH=N), 11.18 (br s, 1H, CONH, disappears with D2O).  $^{13}\text{C-NMR}$  (101 MHz, DMSO-d<sub>6</sub>):  $\delta$  160.84, 154.76, 150.54, 145.70, 144.69, 140.17, 135.25, 134.08, 129.75, 128.64, 128.58, 127.85, 127.72, 126.02, 121.64, 114.35, 109.98, 83.28, 71.29, 67.86, 56.63, 56.02. Elemental Analysis calculated for  $C_{27}H_{24}\text{CIN}_5\text{O}_3$ .

### Antiproliferative activity

To perform MTT assay, SK-BR3 (breast adenocarcinoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy), MCF-7 (breast adenocarcinoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy), SK-Mel28 (skin melanoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy), Hep-G2 (hepatocellular carcinoma, ATCC, Manassas, VA), SKOV-3 (ovarian adenocarcinoma, ATCC, Manassas, VA), MDA-MB231 (breast adenocarcinoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy), HeLa (cervical adenocarcinoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy), A549 (lung carcinoma, Biologic Bank and Cell Factory, IRCCS Policlinico San Martino, Genoa, Italy) and GM-6114 (embryonic human fibroblast, ATCC, Manassas, VA) cell lines were cultured in Dulbecco's Modified Eagle Medium (DMEM), added with 10% Fetal bovine serum (FBS), 2 mM Glutamine and 1% pen-strep. Reagents were acquired from EuroClone (Milan, Italy) and incubated in a humidified environment at 37 °C with 5 % CO<sub>2</sub>. All chemical compounds were dissolved in DMSO to give a 10 mM stock solution. Then, after an intermediate dilution in growth medium, they were added to the cultured cells at a final working concentration of 10  $\mu$ M (1, 5, 10  $\mu$ M for IC $_{50}$  assay) and incubated for 48 hours. At the end of the incubation, 30  $\mu$ L of MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide) at a concentration of 2 mg/mL in phosphate buffered saline (PBS), were added in each well and incubated 4 hours. Finally, the supernatants were removed and 100  $\mu$ L/well of DMSO were added to each well to dissolve the Formazan precipitates. After 20 min, the results were read at 570 nm. Results are expressed as percentage of the control samples, where cells have been treated with the same amount of DMSO but without any chemical compound. The assay was repeated three times and a single compound was tested six times. Means and standard deviations were calculated.

The  $IC_{50}$  values were calculated based on single concentration-response curves. Each experiment was repeated three times.

#### Blood collection and preparative procedures

Freshly drawn venous blood from healthy volunteers of the "Centro Trasfusionale, Ospedale San Martino" in Genoa was collected into 130 mM aqueous trisodium citrate anticoagulant solution (9:1). The donors claimed to have not taken drugs known to interfere with platelet function during two weeks prior to blood collection and gave their in-formed consent. Washed platelets were prepared centrifuging whole blood at  $100\times g$  for 20 min. The obtained platelet-rich plasma was then centrifuged at  $1100\times g$  for 15 min. Pellet was washed once with pH 5.2 ACD solution (75 mM trisodium citrate, 42 mM citric acid and 136 mM glucose), centrifuged at  $1100\times g$  for 15 min and then resuspended in pH 7.4 HEPES buffer (145 mM NaCl, 5 mM KCl, 1 mM MgSO4, 10 mM glucose, 10 mM HEPES).

The reported  $IC_{50}$  values are the molar concentration of the compound able to obtain 50% inhibition of the maximal effect induced by thrombin (Sigma-Aldrich/Merck Millipore) and are calculated by the percentage of inhibition that are the inhibition of the maximal effect measured in the presence of the agent compared with that measured in a control sample containing saline, carried out under the same conditions.

#### **ROS** assay

ROS production was quantified as previously reported  $^{[9,15]}$  by 2',7'-Dichlorofluorescein diacetate (DCFH—DA, Sigma-Aldrich/Merck Millipore), a ROS-sensitive probe that yields, upon oxidation, the fluorescent adduct DCF that is trapped in-side the cells. Briefly, washed platelets ( $1.0\times10^8$ /mL), preincubated with saline and tested compounds for 15 min at 37°C, were stimulated by 0.1 U/mL thrombin. Incubation was stopped by cooling samples in ice bath and then samples were immediately analysed in a Merck Millipore Bioscience Guava easyCyte flow cytometer (Merk Millipore, Burlington, MA, USA). The IC<sub>50</sub> values were calculated as above detailed.

#### Platelet aggregation

Platelet aggregation was performed in a Bio-Data Aggregometer (Bio-Data Corporation, Horsham, PA, USA) according to Born's method  $^{[40]}$  and quantified by the light transmission reached within 6 min at 37 °C. Briefly, washed platelets (3.0×10 $^8$ /mL) were preincubated for 3 min at 37 °C with saline or tested compounds before the addition of 0.1 U/mL thrombin. The IC $_{50}$  values were calculated as above detailed.



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# **Docking studies**

The molecular structures of compounds R-3e, S-3e, R-5h and S-5h were built by MOE2009.10 (builder module), parameterized by MMFF94x force field and their docking poses within tubulin were calculated by AutoDock 4.2.[32] After the removal of water molecules and ligand atoms from the crystal structure of tubulin  $\alpha/\text{tubulin}~\beta/$ stathmin4 (PDB ID: 6XER),[31] polar hydrogen and Gasteiger-Hückel charges were added. The ligands root was defined automatically. A 60×60×60 Å grid (grid spacing 0.375 Å) was centred in the binding site of colchicine, and electrostatic and affinity maps for each atom type of the ligand were calculated. The docking search was performed over 100 conformers using the Genetic Algorithm Local Search protocol as implemented in AutoDock (population size: 50; rate of gene mutation: 0.02; rate of crossover: 0.8). The docking poses were clustered (rmsd: 2.0 Å) and the best conformation of the low energy highest populated cluster was selected as the binding conformation. Model analysis was carried out using the CCP4 program suite.[41] The calculations were run on a Linux PC (Intel® processor Core™ i7-2600 CPU@3.40 GHz).

#### Molecular dynamics studies

The coordinates of S-3e docking pose were parametrized by CHARMM-GUI.[42] The molecular structure of tubulin dimer was processed and parametrized by VMD package (Automatic PSF building module). [43] The protein-ligand complex was assembled and a solvation box including the whole complex (max box padding: x=7; y=7, z=7) was defined. 15059 water molecules were added, and the complex was neutralized by the addition of NaCl ions.

The molecular dynamic simulation was carried out by software NAMD, developed by the Theoretical and Computational Biophysics Group in the Beckman Institute for Advanced Science and Technology at the University of Illinois at Urbana-Champaign. [44] The following parameter were adopted: run = 501000; time step = 2; temperature = 310; restartfreq = 50. VMD "RMSD trajectory tool" and "Hydrogen bonds" were used to analyse the trajectory file. The calculations were run on a Linux PC (Intel® processor Core™ i7-2600 CPU@3.40 GHz).

### **Supporting Information**

The elemental analysis values, cell growth percent values of compounds 3a, 3e, 4c, 5g and 5h on human embryonic fibroblasts (GM-6114) at 10 μM concentration. Predicted pharmacokinetics and drug-like properties of compounds 3a, 3e, 4c, 5g, 5h and the NMR scans of all compounds 3-5 are provided.

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# **Conflict of Interests**

The authors declare no conflict of interest.

# **Data Availability Statement**

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords: imidazo-pyrazoles · anticancer activity · ROS · molecular docking · medicinal chemistry

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